

ALPHA CHEMICALS PTY LTD

Chemwatch Hazard Alert Code: 3

Chemwatch: 2398

Version No: 8.1

Safety Data Sheet according to Work Health and Safety Regulations (Hazardous Chemicals) 2023 and ADG requirements

Issue Date: **20/06/2022** Print Date: **07/11/2024** S.GHS.AUS.EN

SECTION 1 Identification of the substance / mixture and of the company / undertaking

Product Identifier

Product name	TERTIARY BUTYL METHYL ETHER	
Chemical Name	Not Available	
Synonyms	C5-H12-O; (CH3)3COCH3; methyl t-butyl ether; 2-methoxy-2-methylpropane; tert-butyl methyl ether; methyl 1,1-dimethyl ether; propane, 2-methoxy-2-methyl-; MTBE; 2,2-MMOP	
Proper shipping name	METHYL tert-BUTYL ETHER	
Chemical formula	C5-H12-O	
Other means of identification	Not Available	
CAS number	1634-04-4	

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses	Octane booster for unleaded petrol (up to 7% by volume), this use has been banned in California by year 2003, because of ground water contamination. Manufacture of isobutene; solvent for a number of applications, including pesticide analysis.
Relevant identified uses	The use of a quantity of material in an unventilated or confined space may result in increased exposure and an irritating atmosphere developing. Before starting consider control of exposure by mechanical ventilation.

Details of the manufacturer or supplier of the safety data sheet

Registered company name	ALPHA CHEMICALS PTY LTD	
Address	4 ALLEN PLACE WETHERILL PARK NSW 2164 Australia	
Telephone	1 (0)2 9982 4622	
Fax	Not Available	
Website	~	
Email	shane@alphachem.com.au	

Emergency telephone number

Association / Organisation	ALPHA CHEMICALS PTY LTD	CHEMWATCH EMERGENCY RESPONSE (24/7)	
Emergency telephone number(s)	61 (0)418 237 771	+61 1800 951 288	
Other emergency telephone number(s)	Not Available	+61 3 9573 3188	

Once connected and if the message is not in your preferred language then please dial 01

SECTION 2 Hazards identification

Classification of the substance or mixture

HAZARDOUS CHEMICAL. DANGEROUS GOODS. According to the WHS Regulations and the ADG Code.

Chemwatch Hazard Ratings

	Min	Max	
Flammability	3		
Toxicity	1		0 = Minimum
Body Contact	2		1 = Low
Reactivity	1		2 = Moderate
Chronic	1		3 = High 4 = Extreme

Poisons Schedule	Not Applicable
Classification ^[1]	Flammable Liquids Category 2, Skin Corrosion/Irritation Category 2, Serious Eye Damage/Eye Irritation Category 2B
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI

Hazard pictogram(s)	

Signal word Danger

Hazard statement(s)

H225	Highly flammable liquid and vapour.	
H315	Causes skin irritation.	
H320	Causes eye irritation.	

Precautionary statement(s) Prevention

Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking.	
Keep container tightly closed.	
Ground and bond container and receiving equipment.	
Use explosion-proof electrical/ventilating/lighting/intrinsically safe equipment.	
Use non-sparking tools.	
Take action to prevent static discharges.	
80 Wear protective gloves and protective clothing.	
Wash all exposed external body areas thoroughly after handling.	

Precautionary statement(s) Response

P370+P378	In case of fire: Use alcohol resistant foam or normal protein foam to extinguish.	
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.	
P337+P313	If eye irritation persists: Get medical advice/attention.	
P302+P352	IF ON SKIN: Wash with plenty of water.	
P303+P361+P353	IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water [or shower].	
P332+P313	If skin irritation occurs: Get medical advice/attention.	
P362+P364	Take off contaminated clothing and wash it before reuse.	

Precautionary statement(s) Disposal

Precautionary statement(s) Storage

P501 Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

SECTION 3 Composition / information on ingredients

P403+P235 Store in a well-ventilated place. Keep cool.

Substances

CAS No	%[weight]	Name
1634-04-4	>=95	Tertiary Butyl Methyl Ether
Not Available	<=5	commercial material may contain C5 hydrocarbons
Legend: 1. Classified by Cl	nemwatch; 2. Classification drawr	n from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4. Classification drawn from

C&L; * EU IOELVs available

Mixtures

See section above for composition of Substances

SECTION 4 First aid measures

Description of first aid measures

Eye Contact	 If this product comes in contact with the eyes: Wash out immediately with fresh running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Seek medical attention without delay; if pain persists or recurs seek medical attention. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	 If skin contact occurs: Immediately remove all contaminated clothing, including footwear. Flush skin and hair with running water (and soap if available). Seek medical attention in event of irritation.
Inhalation	 If fumes or combustion products are inhaled remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. Transport to hospital, or doctor.
Ingestion	If swallowed do NOT induce vomiting.

If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration Observe the patient carefully. • Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious. • Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink. Seek medical advice

If spontaneous vomiting appears imminent or occurs, hold patient's head down, lower than their hips to help avoid possible aspiration of vomitus

Indication of any immediate medical attention and special treatment needed

Any material aspirated during vomiting may produce lung injury. Therefore emesis should not be induced mechanically or pharmacologically. Mechanical means should be used if it is considered necessary to evacuate the stomach contents; these include gastric lavage after endotracheal intubation. If spontaneous vomiting has occurred after ingestion, the patient should be monitored for difficult breathing, as adverse effects of aspiration into the lungs may be delayed up to 48 hours. Treat symptomatically.

for lower alkyl ethers:

BASIC TREATMENT

- Establish a patent airway with suction where necessary.
- Watch for signs of respiratory insufficiency and assist ventilation as necessary.
 Administer oxygen by non-rebreather mask at 10 to 15 l/min.
- A low-stimulus environment must be maintained.
- Monitor and treat, where necessary, for shock.
- Anticipate and treat, where necessary, for seizures
- DO NOT use emetics. Where ingestion is suspected rinse mouth and give up to 200 ml water (5 ml/kg recommended) for dilution where patient is able to swallow, has a strong gag reflex and does not drool.

ADVANCED TREATMENT

- Consider orotracheal or nasotracheal intubation for airway control in unconscious patient or where respiratory arrest has occurred.
- Positive-pressure ventilation using a bag-valve mask might be of use
- Monitor and treat, where necessary, for arrhythmias.
- Start an IV D5W TKO. If signs of hypovolaemia are present use lactated Ringers solution. Fluid overload might create complications.
- Drug therapy should be considered for pulmonary oedema.
- Hypotension without signs of hypovolaemia may require vasopressors.
- Treat seizures with diazepam.
- Proparacaine hydrochloride should be used to assist eye irrigation.

EMERGENCY DEPARTMENT

- Laboratory analysis of complete blood count, serum electrolytes, BUN, creatinine, glucose, urinalysis, baseline for serum aminotransferases (ALT and AST), calcium, phosphorus and magnesium, may assist in establishing a treatment regime. Other useful analyses include anion and osmolar gaps, arterial blood gases (ABGs), chest radiographs and electrocardiograph.
- Ethers may produce anion gap acidosis. Hyperventilation and bicarbonate therapy might be indicated.
- Haemodialysis might be considered in patients with impaired renal function.

Consult a toxicologist as necessary.

BRONSTEIN, A.C. and CURRANCE, P.L.

EMERGENCY CARE FOR HAZARDOUS MATERIALS EXPOSURE: 2nd Ed. 1994

SECTION 5 Firefighting measures

Extinguishing media

- Foam
- Dry chemical powder. BCF (where regulations permit).
- Carbon dioxide
- Water spray or fog Large fires only.

ocial hazards arising from the substrate or mixture S

Special hazards arising from th	
Fire Incompatibility	• Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result
Advice for firefighters	
Fire Fighting	 Alert Fire Brigade and tell them location and nature of hazard. May be violently or explosively reactive. Wear breathing apparatus plus protective gloves in the event of a fire. Prevent, by any means available, spillage from entering drains or water course. Consider evacuation (or protect in place). Fight fire from a safe distance, with adequate cover. If safe, switch off electrical equipment until vapour fire hazard removed. Use water delivered as a fine spray to control the fire and cool adjacent area.
Fire/Explosion Hazard	 Liquid and vapour are highly flammable. Severe fire hazard when exposed to heat, flame and/or oxidisers. Vapour may travel a considerable distance to source of ignition. Heating may cause expansion or decomposition leading to violent rupture of containers. On combustion, may emit toxic fumes of carbon monoxide (CO). Combustion products include: carbon dioxide (CO2) other pyrolysis products typical of burning organic material. Contains low boiling substance: Closed containers may rupture due to pressure buildup under fire conditions.
HAZCHEM	•3YE

SECTION 6 Accidental release measures

Personal precautions, protective equipment and emergency procedures See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Minor Spills	 Remove all ignition sources. Clean up all spills immediately. Avoid breathing vapours and contact with skin and eyes. Control personal contact with the substance, by using protective equipment. Contain and absorb small quantities with vermiculite or other absorbent material. Wipe up. Collect residues in a flammable waste container.
Major Spills	 Clear area of personnel and move upwind. Alert Fire Brigade and tell them location and nature of hazard. May be violently or explosively reactive. Wear breathing apparatus plus protective gloves. Prevent, by any means available, spillage from entering drains or water course. Consider evacuation (or protect in place). No smoking, naked lights or ignition sources. Increase ventilation.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 Handling and storage

Precautions for safe handling	
Safe handling	 Containers, even those that have been emptied, may contain explosive vapours. Do NOT cut, drill, grind, weld or perform similar operations on or near containers. May form explosive peroxides on standing or following concentration by distillation. Review of stocks and testing for peroxide content by given tested procedures at 3-monthly intervals is recommended, together with safe disposal of peroxidic samples. [Peroxide-containing residues can often be rendered innocuous by pouring into an excess of sodium carbonate solution] Contains low boiling substance: Storage in sealed containers may result in pressure buildup causing violent rupture of containers not rated appropriately. Check for bulging containers. Vent periodically Always release caps or seals slowly to ensure slow dissipation of vapours DO NOT allow clothing wet with material to stay in contact with skin Avoid all personal contact, including inhalation. Wee ar protective clothing when risk of exposure occurs. Use in a well-ventilated area. Prevent concentration in hollows and sumps. DO NOT enter confined spaces until atmosphere has been checked. Avoid smoking, naked lights, heat or ignition sources. When handling, DO NOT eat, drink or smoke. Vapour may ignite on pumping or pouring due to static electricity.
Other information	 Generally, all technical requirements and operational practices which apply for gasoline are applicable for neat ethyl tert-butyl ether and methyl tert-butyl ether (ETBE and MTBE) and also gasolines containing them. However with the following areas requiring specific attention: Gasket materials compatibility to be checked Vapour recovery design, capacity to be checked Storage tanks should have floating roofs and domes Tank bottom water phases to be directed to adapted waste water treatment Special emphasis on leak prevention/detection and soli/groundwater protection Emergency response: oxygenates-compatible extinction foams, adsorbents Minimised response time for soli/groundwater remediation in case of a leakage or splash Rotate all stock to prevent ageing. Use on FIFO (First In-First Out) basis Store below 38 deg. C. Store in original containers in approved flame-proof area. No smoking, naked lights, heat or ignition sources. DO NOT store in pils, depression, basement or areas where vapours may be trapped. Keep containers securely sealed. Store away from incompatible materials in a cool, dry well ventilated area. Protect containers against physical damage and check regularly for leaks. Observe manufacturer's storage and handling recommendations contained within this SDS. Tank storage: Tanks must be specifically designed for use with this product.

Conditions for safe storage, including any incompatibilities

Conditions for safe storage, in	cluding any incompatibilities
Suitable container	 Packing as supplied by manufacturer. Plastic containers may only be used if approved for flammable liquid. Check that containers are clearly labelled and free from leaks. For low viscosity materials (i) : Drums and jerry cans must be of the non-removable head type. (ii) : Where a can is to be used as an inner package, the can must have a screwed enclosure. For materials with a viscosity of at least 2680 cSt. (23 deg. C) For manufactured product having a viscosity of at least 250 cSt. (23 deg. C) Manufactured product that requires stirring before use and having a viscosity of at least 20 cSt (25 deg. C): (i) Removable head packaging; (ii) Cans with friction closures and (iii) low pressure tubes and cartridges may be used. Where combination packages are used, and the inner packages are of glass, there must be sufficient inert cushioning material in contact with inner and outer packagings are glass and contain liquids of packing group I there must be sufficient inert absorben to absorb any spillage, unless the outer packaging is a close fitting moulded plastic box and the substances are not incompatible with the plastic.
Storage incompatibility	Methyl tert-butyl ether (MTBE): - Unlike most ethers, MTBE does not tend to form peroxides (auto-oxidise) during storage (by contrast ethyl tert-butyl ether (ETBE) can form peroxides in the presence of oxygen gas or oxidative agents. The peroxidation reaction is slow in typical storage conditions, but it is enhanced by the UV light) [EFOA - European Fuel Oxygenates Association] - may act as a base- although an ether, MTBE is a poor Lewis base and does not support formation of Grignard reagents - unstable toward strong acids. It reacts dangerously with bromine - reacts violently with oxidisers, permanganates, peroxides and hydroperoxides, ammonium persulfate, bromine dioxide, strong acids, sulfuric acid, nitric acid, acyl halides, Lewis acids (aluminium chloride, boron trifluoride, ferric chloride etc.)

- may generate electrostatic charges due to low conductivity
Ethers
may react violently with strong oxidising agents and acids.
can act as bases they form salts with strong acids and addition complexes with Lewis acids; the complex between diethyl ether and boron
trifluoride is an example.
are generally stable to water under neutral conditions and ambient temperatures.
are hydrolysed by heating in the presence of halogen acids, particularly hydrogen iodide
are relatively inert In other reactions, which typically involve the breaking of the carbon-oxygen bond
The tendency of many ethers to form explosive peroxides is well documented.
Ethers lacking non-methyl hydrogen atoms adjacent to the ether link are thought to be relatively safe.
When solvents have been freed from peroxides (by percolation through a column of activated alumina for example), the absorbed peroxides must promptly be desorbed by treatment with the polar solvents methanol or water, which should be discarded safely.

SECTION 8 Exposure controls / personal protection

Control parameters

Occupational Exposure Limits (OEL)

INGREDIENT DATA							
Source	Ingredient	Material name	TWA		STEL	Peak	Notes
Australia Exposure Standards	Tertiary Butyl Methyl Ether	Methyl-tert butyl ether	25 ppm / 92	2 mg/m3	275 mg/m3 / 75 ppm	Not Available	Not Available
Ingredient	Original IDLH			Revised	IDLH		
Tertiary Butyl Methyl Ether	Not Available		Not Avai	lable			

Exposure controls

Exposure controls	
Appropriate engineering controls	Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is done to reduce the risk. Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly. The design of a ventilation system must match the particular process and chemical or contaminant in use. Employers may need to use multiple types of controls to prevent employee overexposure. For flammable liquids and flammable gases, local exhaust ventilation or a process enclosure ventilation system may be required.
Individual protection measures, such as personal protective equipment	
Eye and face protection	 Safety glasses with side shields. Chemical goggles. [AS/NZS 1337.1, EN166 or national equivalent] Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly.
Skin protection	See Hand protection below
Hands/feet protection	 Wear chemical protective gloves, e.g. PVC. Wear safety footwear or safety gumboots, e.g. Rubber The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application. The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice. Personal hygiene is a key element of effective hand care. Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended. Suitability and durability of glove type is dependent on usage.
Body protection	See Other protection below
Other protection	 Overalls. PVC Apron. PVC protective suit may be required if exposure severe. Eyewash unit. Ensure there is ready access to a safety shower. Some plastic personal protective equipment (PPE) (e.g. gloves, aprons, overshoes) are not recommended as they may produce static electricity. For large scale or continuous use wear tight-weave non-static clothing (no metallic fasteners, cuffs or pockets). Non sparking safety or conductive footwear should be considered. Conductive footwear describes a boot or shoe with a sole made from a conductive compound chemically bound to the bottom components, for permanent control to electrical resistance must range between 0 to 500,000 ohms. Conductive should be stored in lockers close to the room in which they are worn. Personnel who have been issued conductive footwear should not wear them from their place of work to their homes and return.

Respiratory protection

Type AX Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required. Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 5 x ES	AX-AUS / Class 1	-	AX-PAPR-AUS / Class 1
up to 25 x ES	Air-line*	AX-2	AX-PAPR-2

up to 50 x ES	-	AX-3	-
50+ x ES	-	Air-line**	-
^ - Full-face			

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

- Cartridge respirators should never be used for emergency ingress or in areas of unknown vapour concentrations or oxygen content.
- The wearer must be warned to leave the contaminated area immediately on detecting any odours through the respirator. The odour may indicate that the mask is not functioning properly, that the vapour concentration is too high, or that the mask is not properly fitted. Because of these limitations, only restricted use of cartridge respirators is considered appropriate.
- Cartridge performance is affected by humidity. Cartridges should be changed after 2 hr of continuous use unless it is determined that the humidity is less than 75%, in which case, cartridges can be used for 4 hr. Used cartridges should be discarded daily, regardless of the length of time used

Selection of the Class and Type of respirator will depend upon the level of breathing zone contaminant and the chemical nature of the contaminant. Protection Factors (defined as the ratio of contaminant outside and inside the mask) may also be important.

Required minimum protection factor	Maximum gas/vapour concentration present in air p.p.m. (by volume)	Half-face Respirator	Full-Face Respirator
up to 10	1000	AX-AUS / Class 1	-
up to 50	1000	-	AX-AUS / Class 1
up to 50	5000	Airline *	-
up to 100	5000	-	AX-2
up to 100	10000	-	AX-3
100+		-	Airline**

** - Continuous-flow or positive pressure demand.

A(All classes) = Organic vapours, B AUS or B1 = Acid gases, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 deg C)

SECTION 9 Physical and chemical properties

Information on basic physical and chemical properties

Appearance Colourless, highly volatile, highly flammable liquid; ethereal odour; partly mixes with water (4.8% at 25 deg C.)..

Physical state	Liquid	Relative density (Water = 1)	0.74
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	192
pH (as supplied)	Not Applicable	Decomposition temperature (°C)	Not Available
Melting point / freezing point (°C)	-110	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	54-56	Molecular weight (g/mol)	88.15
Flash point (°C)	-28	Taste	Not Available
Evaporation rate	Fast	Explosive properties	Not Available
Flammability	HIGHLY FLAMMABLE.	Oxidising properties	Not Available
Upper Explosive Limit (%)	15.1	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	1.6	Volatile Component (%vol)	100
Vapour pressure (kPa)	27.5 @ 20C	Gas group	Not Available
Solubility in water	Partly miscible	pH as a solution (1%)	Not Applicable
Vapour density (Air = 1)	3.1	VOC g/L	703.74
Heat of Combustion (kJ/g)	Not Available	Ignition Distance (cm)	Not Available
Flame Height (cm)	Not Available	Flame Duration (s)	Not Available
Enclosed Space Ignition Time Equivalent (s/m3)	Not Available	Enclosed Space Ignition Deflagration Density (g/m3)	Not Available

SECTION 10 Stability and reactivity

Reactivity	See section 7
Chemical stability	 Unstable in the presence of incompatible materials. Product is considered stable. Hazardous polymerisation will not occur.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 Toxicological information

Inhaled	Inhalation of vapours may cause drowsiness and dizziness. The lack of co-ordination, and vertigo. There is some evidence to suggest that the material can cause can cause further lung damage. Rats exposed to 8000 ppm MTBE for 6 hours showed change transient central nervous system depression. Transient increas 800 and 4000 ppm. This finding may reflect an exposure-relate anaesthetic effect. Rats exposed to 1000 ppm MTBE, 6 hours/day, 5 days/week f changes. A reduced reaction to auditory stimuli was noted at 3 chronic inflammatory changes in the nasal mucosa and trache Following inhalation, ethers cause lethargy and stupor. Inhalin seizures and possible coma. Inhalation of aerosols (mists, fumes), generated by the materia individual. If exposure to highly concentrated vapour atmosphere is prolor resuscitated - death.	se respiratory irritation in som es in motor activity with the p uses in motor activity, on the o ted stimulant effect or an exa for 9 exposures showed lach 3000 ppm. There were no oth ea ng lower alkyl ethers results in ial during the course of norma	e persons. The body's response to such irritation attern and time course of effects being indicative of other hand, were observed in male rats exposed at ggerated response following recovery from an rymation, conjunctival swelling and corneal her macroscopic pathology findings other than in headache, dizziness, weakness, blurred vision, al handling, may be damaging to the health of the
Ingestion	Swallowing of the liquid may cause aspiration into the lungs w (ICSC13733) Accidental ingestion of the material may be damaging to the h Ingestion of alkyl ethers may produce stupor, blurred vision, he and asphyxia may result.	nealth of the individual.	
Skin Contact	This material can cause inflammation of the skin on contact in The material may accentuate any pre-existing dermatitis condi Skin contact is not thought to have harmful health effects (as of following entry through wounds, lesions or abrasions. Repeated exposure may cause skin cracking, flaking or drying Alkyl ethers may defat and dehydrate the skin producing derm system depression. Open cuts, abraded or irritated skin should not be exposed to i Entry into the blood-stream, through, for example, cuts, abrasis skin prior to the use of the material and ensure that any extern	dition classified under EC Directive g following normal handling a natoses. Absorption may prod this material sions or lesions, may produce	Ind use. Juce headache, dizziness, and central nervous systemic injury with harmful effects. Examine the
		se eve irritation and damage	in some persons.
Eye	There is some evidence to suggest that this material can caus Eye contact with alkyl ethers (vapour or liquid) may produce in		
Eye Chronic		rritation, redness and tears. ancer or mutations but there i may cause some concern fol ne liver, kidneys, and adrenal opmental toxicity only at leve hours of the testes, kidney, liv	s not enough data to make an assessment. lowing repeated or long-term occupational gland and cause mild changes to blood counts. It Is that are toxic to the mother. MTBE does cause er, uterus and blood cell cancers, and should be
	Eye contact with alkyl ethers (vapour or liquid) may produce in There has been some concern that this material can cause ca Substance accumulation, in the human body, may occur and n exposure. Animal testing shows that MTBE can increase the weight of th is not clear whether MTBE causes mutations. It causes develo kidney damage. In animals it has been known to increase tum considered as potentially causing cancer in humans.	rritation, redness and tears. ancer or mutations but there i may cause some concern fol ne liver, kidneys, and adrenal opmental toxicity only at leve hours of the testes, kidney, liv	s not enough data to make an assessment. lowing repeated or long-term occupational gland and cause mild changes to blood counts. It Is that are toxic to the mother. MTBE does cause er, uterus and blood cell cancers, and should be
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	Eye contact with alkyl ethers (vapour or liquid) may produce in There has been some concern that this material can cause ca Substance accumulation, in the human body, may occur and n exposure. Animal testing shows that MTBE can increase the weight of th is not clear whether MTBE causes mutations. It causes develoc kidney damage. In animals it has been known to increase turn considered as potentially causing cancer in humans. Chronic exposure to alkyl ethers may result in loss of appetite, TOXICITY dermal (rat) LD50: >2000 mg/kg ^[1]	rritation, redness and tears. ancer or mutations but there i may cause some concern fol ne liver, kidneys, and adrenal opmental toxicity only at leve hours of the testes, kidney, liv e, excessive thirst, fatigue, an IRRITATION Eye: no adverse of	s not enough data to make an assessment. lowing repeated or long-term occupational gland and cause mild changes to blood counts. It Is that are toxic to the mother. MTBE does cause er, uterus and blood cell cancers, and should be d weight loss.
Chronic TERTIARY BUTYL METHYL	Eye contact with alkyl ethers (vapour or liquid) may produce in There has been some concern that this material can cause ca Substance accumulation, in the human body, may occur and n exposure. Animal testing shows that MTBE can increase the weight of th is not clear whether MTBE causes mutations. It causes develoc kidney damage. In animals it has been known to increase tum considered as potentially causing cancer in humans. Chronic exposure to alkyl ethers may result in loss of appetite, TOXICITY	rritation, redness and tears. ancer or mutations but there i may cause some concern fol opmental toxicity only at leve nours of the testes, kidney, liv e, excessive thirst, fatigue, an IRRITATION Eye: no adverse of Skin: adverse effe	s not enough data to make an assessment. lowing repeated or long-term occupational gland and cause mild changes to blood counts. It Is that are toxic to the mother. MTBE does cause er, uterus and blood cell cancers, and should be d weight loss.
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Chronic TERTIARY BUTYL METHYL ETHER	Eye contact with alkyl ethers (vapour or liquid) may produce in There has been some concern that this material can cause ca Substance accumulation, in the human body, may occur and n exposure. Animal testing shows that MTBE can increase the weight of th is not clear whether MTBE causes mutations. It causes develoc kidney damage. In animals it has been known to increase turn considered as potentially causing cancer in humans. Chronic exposure to alkyl ethers may result in loss of appetite, TOXICITY dermal (rat) LD50: >2000 mg/kg ^[1] Inhalation (Rat) LC50: 41 mg/L4h ^[2] Oral (Rat) LD50: >2000 mg/kg ^[2] 1. Value obtained from Europe ECHA Registered Substances	rritation, redness and tears. ancer or mutations but there i may cause some concern fol the liver, kidneys, and adrenal opmental toxicity only at leve hours of the testes, kidney, live and the testes, and the testes, and the testing at concentrations of 5 these, irritation of the eye, nose there MTBE-blended gasoline and to dissolve gallstones; ir allbladder lining. MTBE may chemosis and conjunctival re	s not enough data to make an assessment. lowing repeated or long-term occupational gland and cause mild changes to blood counts. It Is that are toxic to the mother. MTBE does cause er, uterus and blood cell cancers, and should be d weight loss. effect observed (not irritating) ^[1] ect observed (irritating) ^[1] effect observed (not irritating) ^[1] effect observed (not irritating) ^[1] ained from manufacturer's SDS. Unless otherwise excepte may be adversely affected by volatile oxicity, and the main affected system is the mg/m3 for 1 hour. After the introduction of MTBE- e and throat, cough, nausea, dizziness and e was also used. If symptoms appeared, they were jection of MTBE into the gallbladder can cause also cause mild skin irritation (redness and
Chronic TERTIARY BUTYL METHYL ETHER <i>Legend:</i>	Eye contact with alkyl ethers (vapour or liquid) may produce in There has been some concern that this material can cause ca Substance accumulation, in the human body, may occur and nexposure. Animal testing shows that MTBE can increase the weight of th is not clear whether MTBE causes mutations. It causes develoc kidney damage. In animals it has been known to increase turn considered as potentially causing cancer in humans. Chronic exposure to alkyl ethers may result in loss of appetite, TOXICITY dermal (rat) LD50: >2000 mg/kg ^[1] Inhalation (Rat) LC50: 41 mg/L4h ^[2] Oral (Rat) LD50: >2000 mg/kg ^[2] 1. Value obtained from Europe ECHA Registered Substances specified data extracted from RTECS - Register of Toxic Effect For methyl tert-butyl ether (MTBE): In particular climates (such emissions from MTBE-blended gasoline. Animal testing shows nervous system. In humans, no symptoms were observed in to blended gasoline in Alaska, there were complaints of headach spaciness. This did not occur in New Jersey or Connecticut wf apparently mild and transient. MTBE has been used as treatm nausea, vomiting, sleepiness and temporary damage to the ga swelling) and moderate eye irritation (clouding of the cornea, c The substance is classified by IARC as Group 3: NOT classifiable as to its carcinogenicity to humans.	rritation, redness and tears. ancer or mutations but there i may cause some concern fol the liver, kidneys, and adrenal opmental toxicity only at leve hours of the testes, kidney, live and the testes, and the testes, and the testing at concentrations of 5 these, irritation of the eye, nose there MTBE-blended gasoline and to dissolve gallstones; ir allbladder lining. MTBE may chemosis and conjunctival re	s not enough data to make an assessment. lowing repeated or long-term occupational gland and cause mild changes to blood counts. It Is that are toxic to the mother. MTBE does cause er, uterus and blood cell cancers, and should be d weight loss. effect observed (not irritating) ^[1] ect observed (irritating) ^[1] effect observed (not irritating) ^[1] effect observed (not irritating) ^[1] ained from manufacturer's SDS. Unless otherwise excepte may be adversely affected by volatile oxicity, and the main affected system is the mg/m3 for 1 hour. After the introduction of MTBE- e and throat, cough, nausea, dizziness and e was also used. If symptoms appeared, they were jection of MTBE into the gallbladder can cause also cause mild skin irritation (redness and
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Chronic TERTIARY BUTYL METHYL ETHER Legend: TERTIARY BUTYL METHYL ETHER	Eye contact with alkyl ethers (vapour or liquid) may produce in There has been some concern that this material can cause ca Substance accumulation, in the human body, may occur and nexposure. Animal testing shows that MTBE can increase the weight of the is not clear whether MTBE causes mutations. It causes develoce kidney damage. In animals it has been known to increase turn considered as potentially causing cancer in humans. Chronic exposure to alkyl ethers may result in loss of appetite, TOXICITY dermal (rat) LD50: >2000 mg/kg ^[1] Inhalation (Rat) LC50: 41 mg/L4h ^[2] Oral (Rat) LD50: >2000 mg/kg ^[2] 1. Value obtained from Europe ECHA Registered Substances specified data extracted from RTECS - Register of Toxic Effect For methyl tert-butyl ether (MTBE): In particular climates (such emissions from MTBE-blended gasoline. Animal testing shows nervous system. In humans, no symptoms were observed in the blended gasoline in Alaska, there were complaints of headach spaciness. This did not occur in New Jersey or Connecticut wit apparently mild and transient. MTBE has been used as treatm nausea, vomiting, sleepiness and temporary damage to the ga swelling) and moderate eye irritation (clouding of the cornea, c The substance is classified by IARC as Group 3: NOT classifiable as to its carcinogenicity to humans. Evidence of carcinogenicity may be inadequate or limited in ar	rritation, redness and tears. ancer or mutations but there i may cause some concern fol the liver, kidneys, and adrenal opmental toxicity only at leve hours of the testes, kidney, live the excessive thirst, fatigue, and IRRITATION Eye: no adverse of Skin: adverse effect Skin: adverse effect Skin: no adverse s - Acute toxicity 2. Value obtact of chemical Substances h as subarctic), susceptible p is that MTBE has low acute to testing at concentrations of 5 here, irritation of the eye, nose there MTBE-blended gasoline nent to dissolve gallstones; ir allbladder lining. MTBE may chemosis and conjunctival re- minimal testing. Carcinogenicity	s not enough data to make an assessment. lowing repeated or long-term occupational gland and cause mild changes to blood counts. It Is that are toxic to the mother. MTBE does cause er, uterus and blood cell cancers, and should be d weight loss. effect observed (not irritating) ^[1] act observed (intritating) ^[1] effect observed (not irritating) ^[1] anined from manufacturer's SDS. Unless otherwise expople may be adversely affected by volatile boxicity, and the main affected system is the mg/m3 for 1 hour. After the introduction of MTBE- a and throat, cough, nausea, dizziness and a was also used. If symptoms appeared, they were jection of MTBE into the gallbladder can cause also cause mild skin irritation (redness and dness).
Chronic TERTIARY BUTYL METHYL ETHER Legend: TERTIARY BUTYL METHYL ETHER Acute Toxicity Skin Irritation/Corrosion	Eye contact with alkyl ethers (vapour or liquid) may produce in There has been some concern that this material can cause ca Substance accumulation, in the human body, may occur and nexposure. Animal testing shows that MTBE can increase the weight of the is not clear whether MTBE causes mutations. It causes develoc kidney damage. In animals it has been known to increase turn considered as potentially causing cancer in humans. Chronic exposure to alkyl ethers may result in loss of appetite, TOXICITY dermal (rat) LD50: >2000 mg/kg ^[1] Inhalation (Rat) LC50: 41 mg/L4h ^[2] Oral (Rat) LD50: >2000 mg/kg ^[2] 1. Value obtained from Europe ECHA Registered Substances specified data extracted from RTECS - Register of Toxic Effect For methyl tert-butyl ether (MTBE): In particular climates (such emissions from MTBE-blended gasoline. Animal testing shows nervous system. In humans, no symptoms were observed in the blended gasoline in Alaska, there were complaints of headach spaciness. This did not occur in New Jersey or Connecticut wit apparently mild and transient. MTBE has been used as treatm nausea, vomiting, sleepiness and temporary damage to the gas swelling) and moderate eye irritation (clouding of the cornea, cord The substance is classified by IARC as Group 3: NOT classifiable as to its carcinogenicity to humans. Evidence of carcinogenicity may be inadequate or limited in ar	rritation, redness and tears. ancer or mutations but there i may cause some concern fol the liver, kidneys, and adrenal opmental toxicity only at leve hours of the testes, kidney, liv e, excessive thirst, fatigue, and IRRITATION Eye: no adverse of Skin: adverse effer Skin: no adverse effer Skin: no adverse ct of chemical Substances h as subarctic), susceptible p is that MTBE has low acute to testing at concentrations of 50 res, irritation of the eye, nose there MTBE-blended gasoline nent to dissolve gallstones; ir allbladder lining. MTBE may chemosis and conjunctival re- inimal testing. Carcinogenicity Reproductivity	s not enough data to make an assessment. lowing repeated or long-term occupational gland and cause mild changes to blood counts. It is that are toxic to the mother. MTBE does cause er, uterus and blood cell cancers, and should be d weight loss. effect observed (not irritating) ^[1] effect observed (not irritating) ^[1] effect observed (not irritating) ^[1] ained from manufacturer's SDS. Unless otherwise weople may be adversely affected by volatile oxicity, and the main affected system is the mg/m3 for 1 hour. After the introduction of MTBE- and throat, cough, nausea, dizziness and e was also used. If symptoms appeared, they were also cause mild skin irritation (redness and dness).

SECTION 12 Ecological information

Toxicity

TERTIARY BUTYL METHYL ETHER	Endpoint	Test Duration (hr)	Species	Value	Source
	EC50	96h	Algae or other aquatic plants	184mg/l	1
	EC50	72h	Algae or other aquatic plants	>800mg/l	1

	NOEC(ECx)	96h	Crustacea	15mg/l	1
	EC50	48h	Crustacea	>100mg/l	1
	LC50	96h	Fish	187mg/l	1
Legend: Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 4. US Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. MET (Japan) - Bioconcentration Data 8. Vendor Data					

for Methyl Tert-Butyl Ether (MTBE): Log Kow: 0.94-1.24; Koc: 11.2; Kow: 10.96; Half-life (hr) air: 120-144; Half-life (hr) Surface Water: 4; Henry's atm m3/mol: 5.87E-04; BCF: 1-5; Vapor Pressure: 245 mm Hg; Henry's Law Constant: 5.5 x 10-4 atm-m3/mole: BFC: Insignificant.

Environmental Fate: MTBE is added to gasoline to improve air quality; however, it may cause emissions of other pollutants to increase. MTBE is a volatile organic compound (VOC) substance

Atmospheric Fate: The ozone forming potential of MTBE is estimated to be lower than that for most other non-methane components of urban air including alkenes, aldehydes, non-toluene aromatics, and ethene. As a VOC, MTBE can contribute to the formation of photochemical smog in the presence of other VOCs. MTBE is highly volatile and would be expected to volatilize rapidly from soil surfaces or water. Ninety nine percent of MTBE will partition to the air; however, MTBE leaking from underground gasoline storage tanks may not readily reach the atmosphere. MTBE is not expected to persist in the atmosphere due to rapid reaction with hydroxyl radicals. Most ethers are very resistant to hydrolysis, and the rate of cleavage of the carbon-oxygen bond by abiotic processes is expected to be insignificant.

Direct photolysis will not be an important removal process since aliphatic ethers do not absorb light at wavelengths >290 nm DO NOT discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air			
Tertiary Butyl Methyl Ether	HIGH (Half-life = 360 days)	LOW (Half-life = 11.04 days)			
Bioaccumulative potential					
Bioaccumulative potential					
Bioaccumulative potential Ingredient	Bioaccumulation				

Mobility in soil

Ingredient	Mobility
Tertiary Butyl Methyl Ether	LOW (Log KOC = 5.258)

SECTION 13 Disposal considerations

Waste treatment methods	
Product / Packaging disposal	 Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked. A Hierarchy of Controls seems to be common - the user should investigate: Reduction Reuse Recycling Disposal (if all else fails) This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. If it has been contaminated, it may be possible to reclaim the product by filtration, distillation or some other means. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate. DO NOT allow wash water from cleaning or process equipment to enter drains. It may be necessary to collect all wash water for treatment before disposal. In all cases disposal to sever may be subject to local laws and regulations and these should be considered first. Where in doubt contact the responsible authority. Recycle wherever possible. Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified. Dispose of by: burial in a land-fill specifically licensed to accept chemical and / or pharmaceutical wastes or Incineration in a licensed apparatus (after admixture with suitable combustible material). Decontaminate empty containers. Observe all label safeguards until containers are cleaned and destroyed.

SECTION 14 Transport information

Labels Required

Marine Pollutant	NO
HAZCHEM	•3YE

Land transport (ADG)

14.1. UN number or ID number	2398			
14.2. UN proper shipping name	METHYL tert-BUTYL ETHER			
14.3. Transport hazard class(es)	Class 3 Subsidiary Hazard Not Applicable			
14.4. Packing group	11			

....

TERTIARY BUTYL METHYL ETHER

14.5.	Environmental hazard	Not Applicable			
14.6.	Special precautions for user	Special provisions Not Applicable Limited quantity 1 L			
Air tra	insport (ICAO-IATA / DGR)			
14.1.	UN number	2398			
14.2.	UN proper shipping name	Methyl tert-butyl ether			
		ICAO/IATA Class	3		
14.3.	Transport hazard class(es)	ICAO / IATA Subsidiary Hazard	Not Applicable		
01035(03)		ERG Code	3L		

	14.4. Packing group	I			
	14.5. Environmental hazard	Not Applicable			
		Special provisions	Not Applicable		
		Cargo Only Packing Instructions	364		
		Cargo Only Maximum Qty / Pack	60 L		
	14.6. Special precautions for user	Passenger and Cargo Packing Instructions	353		
		Passenger and Cargo Maximum Qty / Pack	5 L		
		Passenger and Cargo Limited Quantity Packing Instructions	Y341		
		Passenger and Cargo Limited Maximum Qty / Pack	1 L		

Sea transport (IMDG-Code / GGVSee)

	,		
14.1. UN number	2398		
14.2. UN proper shipping name	METHYL tert-BUTYL ETHER		
14.3. Transport hazard class(es)	IMDG Class IMDG Subsidiary Haz	3 rard Not Applicable	
14.4. Packing group	П		
14.5 Environmental hazard	Not Applicable		
14.6. Special precautions for user	EMS Number Special provisions Limited Quantities	F-E , S-D Not Applicable	

14.7.1. Transport in bulk according to Annex II of MARPOL and the IBC code

Product name	Pollution Category	Ship Type
Methyl tert-butyl ether	Z	3

14.7.2. Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	ip			
Tertiary Butyl Methyl Ether	Not Available			
14.7.3. Transport in bulk in accordance with the IGC Code				

Product name	Ship Type
Tertiary Butyl Methyl Ether	Not Available

SECTION 15 Regulatory information

Safety, health and environmental regulations / legislation specific for the substance or mixture

Tertiary Butyl Methyl Ether is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals

Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 2

Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 4 Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 5

Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 6

Australian Inventory of Industrial Chemicals (AIIC)

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs - Not Classified as Carcinogenic

Additional Regulatory Information

Not Applicable

National Inventory Status

National Inventory	Status	
Australia - AIIC / Australia Non- Industrial Use	Yes	
Canada - DSL	Yes	
Canada - NDSL	No (Tertiary Butyl Methyl Ether)	
China - IECSC	Yes	
Europe - EINEC / ELINCS / NLP	Yes	
Japan - ENCS	Yes	
Korea - KECI	Yes	
New Zealand - NZIoC	Yes	
Philippines - PICCS	Yes	
USA - TSCA	All chemical substances in this product have been designated as TSCA Inventory 'Active'	
Taiwan - TCSI	Yes	
Mexico - INSQ	Yes	
Vietnam - NCI	Yes	
Russia - FBEPH	Yes	
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.	

SECTION 16 Other information

Revision Date	20/06/2022
Initial Date	12/05/2005

SDS Version Summary

Version	Date of Update	Sections Updated
7.1	04/08/2017	Hazards identification - Classification, Firefighting measures - Fire Fighter (fire/explosion hazard), Handling and storage - Handling Procedure, Handling and storage - Storage (storage requirement)
8.1	20/06/2022	Expiration. Review and Update

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered

Definitions and abbreviations

- PC TWA: Permissible Concentration-Time Weighted Average
- PC STEL: Permissible Concentration-Short Term Exposure Limit
- IARC: International Agency for Research on Cancer
- ACGIH: American Conference of Governmental Industrial Hygienists
- STEL: Short Term Exposure Limit
- TEEL: Temporary Emergency Exposure Limit.
- IDLH: Immediately Dangerous to Life or Health Concentrations
 ES: Exposure Standard
- OSF: Odour Safety Factor
- NOAEL: No Observed Adverse Effect Level
- LOAEL: Lowest Observed Adverse Effect Level
- TLV: Threshold Limit Value
- LOD: Limit Of Detection
- OTV: Odour Threshold Value BCF: BioConcentration Factors
- BEI: Biological Exposure Index
- DNEL: Derived No-Effect Level
- PNEC: Predicted no-effect concentration
- AIIC: Australian Inventory of Industrial Chemicals
- DSL: Domestic Substances List
- NDSL: Non-Domestic Substances List
- IECSC: Inventory of Existing Chemical Substance in China
- EINECS: European INventory of Existing Commercial chemical Substances
 ELINCS: European List of Notified Chemical Substances
- NLP: No-Longer Polymers
- ENCS: Existing and New Chemical Substances Inventory
- KECI: Korea Existing Chemicals Inventory
- NZIoC: New Zealand Inventory of Chemicals
- PICCS: Philippine Inventory of Chemicals and Chemical Substances TSCA: Toxic Substances Control Act
- TCSI: Taiwan Chemical Substance Inventory
- INSQ: Inventario Nacional de Sustancias Químicas
- NCI: National Chemical Inventory
- FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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