

ALPHA CHEMICALS PTY LTD

Chemwatch Hazard Alert Code: 1

Chemwatch: 46102

Version No: 7.1

Safety Data Sheet according to Work Health and Safety Regulations (Hazardous Chemicals) 2023 and ADG requirements

Issue Date: 23/12/2022 Print Date: 07/11/2024 S.GHS.AUS.EN

SECTION 1 Identification of the substance / mixture and of the company / undertaking

Product Identifier

Product name	POTASSIUM CITRATE
Chemical Name	Not Available
Synonyms	anhydrous = C6-H5-K3-O7 monohydrate = C6-H5-K3-O7.H2O C6-H7-K3-O8; Food Additive 332 citric acid tripotassium salt tripotassium citrate; 1,2,3-propanetricarboxylic acid, 2-hydroxy, tripotassium salt
Chemical formula	C6H8O7.H2O.3K C6H8O7.3K
Other means of identification	Not Available
CAS number	866-84-2

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses	In medicine as antacid, in saline drinks, as a buffer stabiliser in foods. As a sequestrant. Therapeutic or pharmacologically-active agent. ATP citrate lyase (ACLY) inhibitor: • ACLY is a key enzyme in cancer metabolism. • ACLY is involved in glucose and lipid metabolism. • Many ACLY inhibitors were developed as anti-cancer agents. ATP-citrate lyase (ACLY) is a cytosolic enzyme upstream of HMG-CoA reductase in the lipid biosynthesis pathway that catalyses the cleavage of mitochondrial-derived citrate into oxaloacetate and acetyl-CoA with the simultaneous hydrolysis of ATP to ADP and phosphate Acetyl-CoA serves as common substrate for de novo cholesterol and fatty acid synthesis. Although ACLY is not rate limiting, its strategic position at the intersection of lipid and carbohydrate metabolism, and its potential to regulate lipoprotein metabolism, attracted early interest as a drug target to treat dyslipidemia. ACLY is an extra-mitochondrial enzyme that is expressed in lipogenic tissues such as liver and white adipose tissues. ACLY is a cytosolic homotetrameric enzyme that catalyzes the conversion of citrate and coenzyme A (CoA) to acetyl-CoA and oxaloacetate, with the simultaneous hydrolysis of ATP to ADP and phosphate.
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Details of the manufacturer or supplier of the safety data sheet

Registered company name	ALPHA CHEMICALS PTY LTD
Address	4 ALLEN PLACE WETHERILL PARK NSW 2164 Australia
Telephone	61 (0)2 9982 4622
Fax	Not Available
Website	~
Email	shane@alphachem.com.au

Emergency telephone number

Association / Organisation	ALPHA CHEMICALS PTY LTD	CHEMWATCH EMERGENCY RESPONSE (24/7)
Emergency telephone number(s)	61 (0)418 237 771	+61 1800 951 288
Other emergency telephone number(s)	Not Available	+61 3 9573 3188

Once connected and if the message is not in your preferred language then please dial 01

SECTION 2 Hazards identification

Classification of the substance or mixture

NON-HAZARDOUS CHEMICAL. NON-DANGEROUS GOODS. According to the WHS Regulations and the ADG Code.

Chemwatch Hazard Ratings

	Min	Max	
Flammability	1 📃		
Toxicity	1		0 = Minimum
Body Contact	0		1 = Low
Reactivity	1		2 = Moderate
Chronic	0		3 = Figh 4 = Extreme

Poisons Schedule	Not Applicable
Classification ^[1]	Non hazardous

Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI
Label elements	
Hazard pictogram(s)	Not Applicable
Signal word	Not Applicable
Hazard statement(s)	
Hazard statement(s)	
Not Applicable	
Precautionary statement(s) Pre	evention
Not Applicable	
Precautionary statement(s) Re	sponse
Not Applicable	
Precautionary statement(s) Sto	prage
Not Applicable	
Precautionary statement(s) Dis	sposal
Not Applicable	

SECTION 3 Composition / information on ingredients

Substances CAS No %[weight] Name 866-84-2 >99 potassium citrate 1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4. Classification drawn from C&L; * EU IOELVs available Legend:

Mixtures

See section above for composition of Substances

SECTION 4 First aid measures

Description of first aid measures		
Eye Contact	 If this product comes in contact with the eyes: Wash out immediately with fresh running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Seek medical attention without delay; if pain persists or recurs seek medical attention. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel. 	
Skin Contact	If skin or hair contact occurs: Flush skin and hair with running water (and soap if available). Seek medical attention in event of irritation. 	
Inhalation	 If dust is inhaled, remove from contaminated area. Encourage patient to blow nose to ensure clear passage of breathing. If irritation or discomfort persists seek medical attention. 	
Ingestion	 If swallowed do NOT induce vomiting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration. Observe the patient carefully. Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious. Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink. Seek medical advice. 	

Indication of any immediate medical attention and special treatment needed

For potassium intoxications:

- Hyperkalaemia, in patients with abnormal renal function, results from reduced renal excretion following intoxication.
- The presence of electrocardiographic evidence of hyperkalemia or serum potassium levels exceeding 7.5 mE/L indicates a medical emergency requiring an intravenous line and constant cardiac monitoring.
- The intravenous ingestion of 5-10 ml of 10% calcium gluconate, in adults, over a 2 minute period antagonises the cardiac and neuromuscular effects. The duration of action is approximately 1 hour. [Ellenhorn and Barceloux: Medical Toxicology]

SECTION 5 Firefighting measures

Extinguishing media

- Foam.
- Dry chemical powder.BCF (where regulations permit).
- Carbon dioxide.
- Water spray or fog Large fires only.

Special hazards arising from the substrate or mixture

Fire Incompatibility	Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result
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Advice for firefighters

Fire Fighting

 Fire/Explosion Hazard Combustible solid which burns but propagates flame with difficulty; it is estimated that most organic dusts are combustible (circa 70%) - according to the circumstances under which the combustion process occurs, such materials may cause fires and / or dust explosions. Organic powders when finely divided over a range of concentrations regardless of particulate size or shape and suspended in air or some other oxidizing medium may form explosive dust-air mixtures and result in a fire or dust explosion (including secondary explosions). Avoid generating dust, particularly clouds of dust in a confined or unventilated space as dusts may form an explosive mixture with air, and any source of ignition, i.e. flame or spark, will cause fire or explosion. Dust clouds generated by the fine grinding of the solid are a particular hazard, accumulations of fine dust (420 micron or less) may burn rapidly and fiercely if ignited - particles exceeding this limit will generally not form flammable dust clouds; once initiated, however, larger particles up to 1400 microns diameter will contribute to the propagation of an explosive limit (LEL) and upper explosive limit (UEL) are applicable to dust clouds but only the LEL is of practical use; - this is because of lower explosive limit (LEL) and upper explosive limit (UEL) are applicable to dust clouds the LEL is of the called the "Minimum Explosible Concentration", MEC). When processed with flammable liquids/vapors/mists.ignitable (hybrid) mixtures may be formed with combustible dusts. Ignitable mixture will be lower than the pure dust in air mixture. The Lower Explosive Limit (LEL) of the vapour/dust mixture will be lower than the pure dust in air mixture. The Lower Explosive Limit (LEL) of the vapour/dust mixture will be lower than the pure dust. Ignitable carbon monoxide (CO) carbon dioxide (CO2) metal oxides May emit poisonous fumes. 		 Wear breathing apparatus plus protective gloves. Prevent, by any means available, spillage from entering drains or water courses. Use water delivered as a fine spray to control fire and cool adjacent area. DO NOT approach containers suspected to be hot. Cool fire exposed containers with water spray from a protected location. If safe to do so, remove containers from path of fire. Equipment should be thoroughly decontaminated after use.
HAZCHEM Not Applicable	Fire/Explosion Hazard	 Combustible solid which burns but propagates flame with difficulty; it is estimated that most organic dusts are combustible (circa 70%) - according to the circumstances under which the combustion process occurs, such materials may cause fires and / or dust explosions. Organic powders when finely divided over a range of concentrations regardless of particulate size or shape and suspended in air or some other oxidizing medium may form explosive dust-air mixtures and result in a fire or dust explosion (including secondary explosions). Avoid generating dust, particularly clouds of dust in a confined or unventilated space as dusts may form an explosive mixture with air, and any source of ignition, i.e. flame or spark, will cause fire or explosion. Dust clouds generated by the fine grinding of the solid are a particular hazard; accumulations of fine dust (420 micron or less) may burn rapidly and fiercely if ignited - particles exceeding this limit will generally not form flammable dust clouds; once initiated, however, larger particles up to 1400 microns diameter will contribute to the propagation of an explosive limit (LEL) and upper explosive limit (UEL) are applicable to dust clouds but only the LEL is of practical use; this is because of the inherent difficulty of achieving homogeneous dust clouds at high temperatures (for dusts the LEL is often called the "Minimum Explosible Concentration", MEC). When processed with flammable liquids/vapors/mists,ignitable (hybrid) mixtures may be formed with combustible dusts. Ignitable mixtures will increase the rate of explosion pressure rise and the Minimum Ignition Energy (the minimum amount of energy required to ignite dust clouds - MIE) will be lower than the pure dust. In air mixture. The Lower Explosive Limit (LEL) of the vapour/dust mixture will be lower than the pure dust. combustion products include: carbon monoxide (CO) carbon monoxide (CO2) metal oxides other pyrolysis products
	HAZCHEM	Not Applicable

SECTION 6 Accidental release measures

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Minor Spills	 Clean up waste regularly and abnormal spills immediately. Avoid breathing dust and contact with skin and eyes. Wear protective clothing, gloves, safety glasses and dust respirator. Use dry clean up procedures and avoid generating dust. Vacuum up or sweep up. NOTE: Vacuum cleaner must be fitted with an exhaust micro filter (H-Class HEPA type) (consider explosion-proof machines designed to be grounded during storage and use). H-Class HEPA filtered industrial vacuum cleaners should NOT be used on wet materials or surfaces. Dampen with water to prevent dusting before sweeping.
Major Spills	 Moderate hazard. CAUTION: Advise personnel in area. Alert Emergency Services and tell them location and nature of hazard. Control personal contact by wearing protective clothing. Prevent, by any means available, spillage from entering drains or water courses. Recover product wherever possible. IF DRY: Use dry clean up procedures and avoid generating dust. Collect residues and place in sealed plastic bags or other containers for disposal.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 Handling and storage

sonal contact, including inhalation. tive clothing when risk of exposure occurs. I-ventilated area. centration in hollows and sumps. ter confined spaces until atmosphere has been checked. by material to contact humans, exposed food or food utensils. ct with incompatible materials. ing, DO NOT eat, drink or smoke. rders when finely divided over a range of concentrations regardless of particulate size or shape and suspended in air or oxidizing medium may form explosive dust-air mixtures and result in a fire or dust explosion (including secondary explosions) borne dust and eliminate all ignition sources. Keep away from heat, hot surfaces, sparks, and flame. od housekeeping practices. st accumulations on a regular basis by vacuuming or gentle sweeping to avoid creating dust clouds. ous suction at points of dust generation to capture and minimise the accumulation of dusts. Particular attention should be rhead and hidden horizontal surfaces to minimise the probability of a "secondary" explosion. According to NFPA Standard yers 1/32 in.(0.8 mm) thick can be sufficient to warrant immediate cleaning of the area. air hoses for cleaning.
e under inert gas. ygroscopic, i.e. absorbs moisture from the air. Keep containers well sealed in storage. inal containers. ners securely sealed. ool, dry area protected from environmental extremes. from incompatible materials and foodstuff containers.

	Protect containers against physical damage and check regularly for leaks.
	 Observe manufacturer's storage and handling recommendations contained within this SDS.
	For major quantities:
	Consider storage in bunded areas - ensure storage areas are isolated from sources of community water (including stormwater, ground water, lakes and streams).
	Ensure that accidental discharge to air or water is the subject of a contingency disaster management plan; this may require consultation with local authorities.
Conditions for safe storage, in	cluding any incompatibilities
	Multi-ply paper bag with sealed plastic liner or heavy gauge plastic bag.

Suitable container	 NOTE: Bags should be stacked, blocked, interlocked, and limited in height so that they are stable and secure against sliding or collapse. Check that all containers are clearly labelled and free from leaks. Packing as recommended by manufacturer. Glass container is suitable for laboratory quantities
Storage incompatibility	Avoid reaction with oxidising agents

SECTION 8 Exposure controls / personal protection

Control parameters

Occupational Exposure Limits (OEL)	
INGREDIENT DATA		
Not Available		
Ingredient	Original IDLH	Revised IDLH
potassium citrate	Not Available	Not Available

Exposure controls

Appropriate engineering controls	Enclosed local exhaust ventilation is required at points of dust, fume or vapour generation. HEPA terminated local exhaust ventilation should be considered at point of generation of dust, fumes or vapours. Barrier protection or laminar flow cabinets should be considered for laboratory scale handling. A fume hood or vented balance enclosure is recommended for weighing/ transferring quantities exceeding 500 mg. When handling quantities up to 500 gram in either a standard laboratory with general dilution ventilation (e.g. 6-12 air changes per hour) is preferred. Quantities up to 1 kilogram may require a designated laboratory using fume hood, biological safety cabinet, or approved vented enclosures. Quantities exceeding 1 kilogram should be handled in a designated laboratory or containment laboratory using appropriate barrier/ containment technology. Manufacturing and pilot plant operations require barrier/ containment and direct coupling technologies.
Individual protection measures, such as personal protective equipment	
Eye and face protection	 When handling very small quantities of the material eye protection may not be required. For laboratory, larger scale or bulk handling or where regular exposure in an occupational setting occurs: Chemical goggles. [AS/NZS 1337.1, EN166 or national equivalent] Face shield. Full face shield may be required for supplementary but never for primary protection of eyes. Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available.
Skin protection	See Hand protection below
Hands/feet protection	 The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application. The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice. Personal hygiene is a key element of effective hand care. Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended. Suitability and durability of glove type is dependent on usage. Rubber gloves (nitrile or low-protein, powder-free latex, latex/ nitrile). Employees allergic to latex gloves should use nitrile gloves in preference. Double gloving should be considered. PVC gloves. Change gloves frequently and when contaminated, punctured or torn. Wash hands immediately after removing gloves. Protective shoe covers. [AS/NZS 2210] Head covering. Experience indicates that the following polymers are suitable as glove materials for protection against undissolved, dry solids, where abrasive particles are not present. polychloroprene. nitrile rubber. butyl rubber. butyl rubber. butyl rubber. butyl rubber. butyl rubber. polyvinyl chloride. Gloves should be examined for wear and/ or degradation constantly.
Body protection	See Other protection below
Other protection	 For quantities up to 500 grams a laboratory coat may be suitable. For quantities up to 1 kilogram a disposable laboratory coat or coverall of low permeability is recommended. Coveralls should be buttoned at collar and cuffs. For quantities over 1 kilogram and manufacturing operations, wear disposable coverall of low permeability and disposable shoe covers. For manufacturing operations, air-supplied full body suits may be required for the provision of advanced respiratory protection. Eye wash unit. Ensure there is ready access to an emergency shower. For Emergencies: Vinyl suit

Respiratory protection

Type -P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	P1 Air-line*	-	PAPR-P1 -
up to 50 x ES	Air-line**	P2	PAPR-P2
up to 100 x ES	-	P3	-
		Air-line*	-
100+ x ES	-	Air-line**	PAPR-P3

* - Negative pressure demand ** - Continuous flow

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

· Respirators may be necessary when engineering and administrative controls do not adequately prevent exposures.

• The decision to use respiratory protection should be based on professional judgment that takes into account toxicity information, exposure measurement data, and frequency and likelihood of the worker's exposure - ensure users are not subject to high thermal loads which may result in heat stress or distress due to personal protective equipment (powered, positive flow, full face apparatus may be an option).

Published occupational exposure limits, where they exist, will assist in determining the adequacy of the selected respiratory protection. These may be government mandated or vendor recommended.

Certified respirators will be useful for protecting workers from inhalation of particulates when properly selected and fit tested as part of a complete respiratory protection program.

• Where protection from nuisance levels of dusts are desired, use type N95 (US) or type P1 (EN143) dust masks. Use respirators and components tested and approved under appropriate government standards such as NIOSH (US) or CEN (EU)

· Use approved positive flow mask if significant quantities of dust becomes airborne.

· Try to avoid creating dust conditions.

SECTION 9 Physical and chemical properties

Information on basic physical and chemical properties

Appearance Colourless or white crystals, granules or powder; soluble in water. Insoluble in alcohol, soluble in glycerol. Mild alkaline reaction. Fresh saline taste. Available in technical, pure and food grades. CAS RN 866-84-2 = anhydrous CAS RN 6100-05-6 = monohydrate Material is hygroscopic, absorbs moisture from surrounding air.

Physical state	Divided Solid	Relative density (Water = 1)	1.98
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Applicable	Decomposition temperature (°C)	270-275
Melting point / freezing point (°C)	275 (dec.)	Viscosity (cSt)	Not Applicable
Initial boiling point and boiling range (°C)	Not Applicable	Molecular weight (g/mol)	324.42
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Applicable	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Applicable
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Negligible
Vapour pressure (kPa)	Negligible	Gas group	Not Available
Solubility in water	Soluble.	pH as a solution (1%)	1 Mol 8.0-9.5
Vapour density (Air = 1)	Not Applicable	VOC g/L	Not Applicable
Heat of Combustion (kJ/g)	Not Available	Ignition Distance (cm)	Not Available
Flame Height (cm)	Not Available	Flame Duration (s)	Not Available
Enclosed Space Ignition Time Equivalent (s/m3)	Not Available	Enclosed Space Ignition Deflagration Density (g/m3)	Not Available

SECTION 10 Stability and reactivity

Reactivity	See section 7
Chemical stability	 Unstable in the presence of incompatible materials. Product is considered stable. Hazardous polymerisation will not occur.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 Toxicological information

Information on toxicological effects

Inhaled	The material is not thought to produce either adverse EC Directives using animal models). Nevertheless, a one other route and good hygiene practice requires to occupational setting. Persons with impaired respiratory function, airway di disability if excessive concentrations of particulate an If prior damage to the circulatory or nervous systems conducted on individuals who may be exposed to fur Not normally a hazard due to non-volatile nature of p	e health effects or irritation of the adverse systemic effects have be that exposure be kept to a mini- iseases and conditions such as re inhaled. Is has occurred or if kidney dam rther risk if handling and use of product	e respiratory tract following inhalation (as classified by een produced following exposure of animals by at least num and that suitable control measures be used in an emphysema or chronic bronchitis, may incur further age has been sustained, proper screenings should be the material result in excessive exposures.
Ingestion	Accidental ingestion of the material may be damagin Use as a food additive indicates good tolerance of so Acute potassium poisoning after swallowing is rare, I weak pulse, irregularities in heart rhythm, heart block	ng to the health of the individual mall amounts, but excessive an because vomiting usually occur k and an eventual fall in blood p	ounts or overuse may bring irritant and/or harmful effects s and renal excretion is fast. Potassium causes a slow, ressure.
Skin Contact	The material is not thought to produce adverse healt models). Nevertheless, good hygiene practice requir occupational setting. Open cuts, abraded or irritated skin should not be ex Entry into the blood-stream, through, for example, cu skin prior to the use of the material and ensure that a	th effects or skin irritation follow res that exposure be kept to a n posed to this material uts, abrasions or lesions, may p any external damage is suitably	ng contact (as classified by EC Directives using animal inimum and that suitable gloves be used in an roduce systemic injury with harmful effects. Examine the protected.
Eye	Although the material is not thought to be an irritant of discomfort characterised by tearing or conjunctival re-	(as classified by EC Directives) edness (as with windburn). Slig	direct contact with the eye may cause transient at abrasive damage may also result.
Chronic	Long-term exposure to the product is not thought to animal models); nevertheless exposure by all routes Long term exposure to high dust concentrations may micron penetrating and remaining in the lung.	produce chronic effects advers should be minimised as a mat / cause changes in lung functio	e to the health (as classified by EC Directives using er of course. n i.e. pneumoconiosis, caused by particles less than 0.5
		IRRITATIC	N
POTASSIUM CITRATE	dermal (rat) LD50: >2000 mg/kg ^[1]	Eye: no ac	verse effect observed (not irritating) ^[1]
	Oral (Mouse) LD50; 5400 mg/kg ¹¹	Skin: no a	verse effect observed (not irritating) ¹¹
Legend:	1. Value obtained from Europe ECHA Registered Su specified data extracted from RTECS - Register of T	ibstances - Acute toxicity 2. Vai Toxic Effect of chemical Substar	e obtained from manufacturer's SDS. Unless otherwise ces
POTASSIUM CITRATE	For HIF ((hypoxia-inducible factor) inhibitors Considering that endothelial HIF-1alpha was shown unintended consequences for ventricular adaptation haemodynamics over a short period, but a detailed fi Under normoxic conditions, HIF-1alpha and HIF-2alg ubiquitinated, and rapidly degraded. PHD activity be induction of HIF activity. Additionally, the observation that mice with loss of PI PHD inhibitors, which are currently in development fi assessments were made based on short-term use. S No significant acute toxicological data identified in lit For citric acid (and its inorganic citrate salts) Based on extensive animal testing data and on hum cancer, birth defects or reproductive toxicity. Further, contrast, irritation, particularly of the eyes but also th	to be critical for left heart adapt in pulmonary hypertension (PF unctional analysis at later time oha are hydroxylated by PHD (p comes rate limited during hypo HD2 developed severe PH sho or chronic anemia. Early clinica Serious pulmonary side effects erature search. an experience, citric acid has lo it does not cause mutations. A e airways and the skin, is the n	ation to overload, systemically targeting HIFs might have). HIF-2 inhibition appeared to improve right ventricular points would be prudent. rolyl hydroxylase domain) proteins (particularly PHD2), ia, allowing accumulation of HIF-1alpha/2alpha and Ild raise a cautionary flag regarding the clinical use of trials did not report any major side effects, but would be possible with chronic use of PHD inhibitors. w acute toxicity. Citric acid is not suspected of causing so, the sensitizing potential is considered low. In ain hazard presented by citric acid.
Acute Toxicity	×	Carcinogenio	ity 🗙
Skin Irritation/Corrosion	×	Reproductiv	ity 🗙
Serious Eye Damage/Irritation	×	STOT - Single Expos	re 🗙
Respiratory or Skin	×	STOT - Repeated Expos	re 🗙

SECTION 12 Ecological information

Mutagenicity

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Toxicity					
	Endpoint	Test Duration (hr)	Species	Value	Source
POTASSIUM CITRATE	NOEC(ECx)	192h	Algae or other aquatic plants	425mg/l	2
Legend:	Extracted from 1 Ecotox database (Japan) - Biocor	. IUCLID Toxicity Data 2. Europe E e - Aquatic Toxicity Data 5. ECETOC acentration Data 8. Vendor Data	CHA Registered Substances - Ecotoxicological Information - A C Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconc	quatic Toxicity entration Data	4. US EPA, 7. METI

Legend:

Aspiration Hazard

×

- Data available to make classification

Data either not available or does not fill the criteria for classification
 Data available to make classification

90citric

DO NOT discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
	No Data available for all ingredients	No Data available for all ingredients

Ingredient	Bioaccumulation
	No Data available for all ingredients
Mobility in soil	
Ingredient	Mobility
	No Data available for all ingredients

SECTION 13 Disposal considerations

Waste treatment methods
Product / Packaging disposal

SECTION 14 Transport information

Labels Required

Marine Pollutant	NO
HAZCHEM	Not Applicable

Land transport (ADG): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

14.7.1. Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

14.7.2. Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
potassium citrate	Not Available

14.7.3. Transport in bulk in accordance with the IGC Code

Product name	Ship Type
potassium citrate	Not Available

SECTION 15 Regulatory information

Safety, health and environmental regulations / legislation specific for the substance or mixture

potassium citrate is found on the following regulatory lists

Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 4 Australian Inventory of Industrial Chemicals (AIIC)

Additional Regulatory Information

Not Applicable

National Inventory Status

National Inventory	Status
Australia - AIIC / Australia Non- Industrial Use	Yes
Canada - DSL	Yes
Canada - NDSL	Yes
China - IECSC	Yes
Europe - EINEC / ELINCS / NLP	Yes
Japan - ENCS	Yes
Korea - KECI	Yes
New Zealand - NZIoC	Yes
Philippines - PICCS	Yes

end of SDS

National Inventory	Status		
USA - TSCA	All chemical substances in this product have been designated as TSCA Inventory 'Active'		
Taiwan - TCSI	Yes		
Mexico - INSQ	Yes		
Vietnam - NCI	Yes		
Russia - FBEPH	Yes		
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.		

SECTION 16 Other information

Revision Date	23/12/2022
Initial Date	23/04/2005

SDS Version Summary

Version	Date of Update	Sections Updated
6.1	23/10/2020	Disposal considerations - Disposal, Exposure controls / personal protection - Engineering Control, Exposure controls / personal protection - Exposure Standard, Exposure controls / personal protection - Personal Protection (other), Exposure controls / personal protection - Personal Protection (other), Exposure controls / personal protection - Personal Protection (other), Exposure controls / personal protection - Personal Protection (eye), Exposure controls / personal protection - Personal Protection (hands/feet), Accidental release measures - Spills (minor), Handling and storage - Storage (storage requirement), Handling and storage - Storage (suitable container), Toxicological information - Toxicity and Irritation (Other), Identification of the substance / mixture and of the company / undertaking - Use
7.1	23/12/2022	Classification review due to GHS Revision change.

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

Definitions and abbreviations

- PC TWA: Permissible Concentration-Time Weighted Average
- PC STEL: Permissible Concentration-Short Term Exposure Limit
- IARC: International Agency for Research on Cancer
- ACGIH: American Conference of Governmental Industrial Hygienists
- STEL: Short Term Exposure Limit
 TEEL: Temporary Emergency Exposure Limit.
- IDLH: Immediately Dangerous to Life or Health Concentrations
- ES: Exposure Standard
- OSF: Odour Safety Factor
- NOAEL: No Observed Adverse Effect Level
- LOAEL: Lowest Observed Adverse Effect Level
- TLV: Threshold Limit Value
- LOD: Limit Of Detection
- OTV: Odour Threshold Value
- BCF: BioConcentration Factors
- BEI: Biological Exposure Index DNEL: Derived No-Effect Level
- PNEC: Predicted no-effect concentration

AIIC: Australian Inventory of Industrial Chemicals

- DSL: Domestic Substances List
- NDSL: Non-Domestic Substances List
- IECSC: Inventory of Existing Chemical Substance in China EINECS: European INventory of Existing Commercial chemical Substances
- ELINCS: European List of Notified Chemical Substances
- NLP: No-Longer Polymers
- ENCS: Existing and New Chemical Substances Inventory
- KECI: Korea Existing Chemicals Inventory
 NZIoC: New Zealand Inventory of Chemicals
- PICCS: Philippine Inventory of Chemicals and Chemical Substances
- TSCA: Toxic Substances Control Act
- TCSI: Taiwan Chemical Substance Inventory
- INSQ: Inventario Nacional de Sustancias Químicas
- NCI: National Chemical Inventory
- FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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