

ALPHA CHEMICALS PTY LTD

Chemwatch Hazard Alert Code: 3

Chemwatch: 10064

Version No: 3.1

Safety Data Sheet according to Work Health and Safety Regulations (Hazardous Chemicals) 2023 and ADG requirements

Issue Date: **20/06/2022** Print Date: **07/11/2024** S.GHS.AUS.EN

SECTION 1 Identification of the substance / mixture and of the company / undertaking

Product Identifier

Product name	CALCIUM HYDROXIDE	
Chemical Name	Not Available	
Synonyms	Ca-H2-O2; Ca(OH)2; calcium hydrate; caustic lime; hydrated lime; slaked lime; lime water; milk of lime; limbux lime; setelime; kemikal; lime hydrated; hydrolime; 27599; Biocalc; Loeschkalk; Calkyl; calcium hydroxide, FCC, low aluminum powder; calcium(II) hydroxide; hydralime; Limbux; Rhenofit CF; calcium hidroxide; white lime; Calbit; Super Microstar; pH12.45 Buffer; calcium hydroxide hydrate *; calcium hydroxide; calcium hydroxide; calcium hydroxide; hydrated; calcium hydroxide UNILAB; calcium hydroxide GPR	
Proper shipping name	CORROSIVE SOLID, BASIC, INORGANIC, N.O.S. (contains calcium hydroxide)	
Chemical formula	Ca(OH)2	
Other means of identification	Not Available	
CAS number	1305-62-0	

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses Laboratory reagent. A large volume industrial chemical. Manufacture of calcium salts. A binder in mortar, plaster, cement and in building and paving materials. A component in drilling muds, pesticides, fireproof coatings, water paints. As an acid neutralizing agent in water and sewage treatment. Disinfectant. As a flux in steel production; in manufacture of paper pulp.

Details of the manufacturer or supplier of the safety data sheet

Registered company name	ALPHA CHEMICALS PTY LTD
Address	4 ALLEN PLACE WETHERILL PARK NSW 2164 Australia
Telephone	61 (0)2 9982 4622
Fax	Not Available
Website	~
Email	shane@alphachem.com.au

Emergency telephone number

Association / Organisation	ALPHA CHEMICALS PTY LTD CHEMWATCH EMERGENCY RESPONSE (24/7)	
Emergency telephone number(s)	61 (0)418 237 771	+61 1800 951 288
Other emergency telephone number(s)	Not Available	+61 3 9573 3188

Once connected and if the message is not in your preferred language then please dial 01

SECTION 2 Hazards identification

Classification of the substance or mixture

HAZARDOUS CHEMICAL. DANGEROUS GOODS. According to the WHS Regulations and the ADG Code.

Chemwatch Hazard Ratings

	Min	Max	
Flammability	0		
Toxicity	0		0 = Minimum
Body Contact	3		1 = Low
Reactivity	0		2 = Moderate
Chronic	0		3 = High 4 = Extreme

Poisons Schedule	Not Applicable	
Classification ^[1]	Corrosive to Metals Category 1, Skin Corrosion/Irritation Category 1B, Serious Eye Damage/Eye Irritation Category 1	
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI	

Hazard pictogram(s)	
Signal word	Danger
Hazard statement(s)	
H290	May be corrosive to metals.
H314	Causes severe skin burns and eye damage.
Precautionary statement(s) Pre	evention
P260	Do not breathe dust/fume.
P264	Wash all exposed external body areas thoroughly after handling.
P280	Wear protective gloves, protective clothing, eye protection and face protection.
P234	Keep only in original packaging.
Precautionary statement(s) Re	sponse
P301+P330+P331	IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.
P303+P361+P353	IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water [or shower].
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
P310	Immediately call a POISON CENTER/doctor/physician/first aider.
P363	Wash contaminated clothing before reuse.
P390	Absorb spillage to prevent material damage.
P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.
Precautionary statement(s) Sto	prage
P405	Store locked up.
Precautionary statement(s) Dis	posal
	Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

SECTION 3 Composition / information on ingredients

Substances			
CAS No		%[weight]	Name
1305-62-0		>95	calcium hydroxide
Legend: 1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4. Classification drawn from C&L * EU IOELVs available			

Mixtures

See section above for composition of Substances

SECTION 4 First aid measures

Eye Contact	 If this product comes in contact with the eyes: Immediately hold eyelids apart and flush the eye continuously with running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes. Transport to hospital or doctor without delay. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	 If skin or hair contact occurs: Immediately flush body and clothes with large amounts of water, using safety shower if available. Quickly remove all contaminated clothing, including footwear. Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre. Transport to hospital, or doctor.
Inhalation	 If fumes or combustion products are inhaled remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures. Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary. Transport to hospital, or doctor. Inhalation of vapours or aerosols (mists, fumes) may cause lung oedema. Corrosive substances may cause lung damage (e.g. lung oedema, fluid in the lungs). As this reaction may be delayed up to 24 hours after exposure, affected individuals need complete rest (preferably in semi-recumbent posture) and must be kept under medical observation even if no symptoms are (yet) manifested. Before any such manifestation, the administration of a spray containing a dexamethasone derivative or beclomethasone derivative may be considered. This must definitely be left to a doctor or person authorised by him/her. (ICSC13719)
Ingestion	For advice, contact a Poisons Information Centre or a doctor at once.

- Urgent hospital treatment is likely to be needed.
- If swallowed do NOT induce vomiting
- F If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.
- Observe the patient carefully.
- Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.
- Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.
 - Transport to hospital or doctor without delay.

Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

- For acute or short-term repeated exposures to highly alkaline materials:
- Respiratory stress is uncommon but present occasionally because of soft tissue edema.
- Unless endotracheal intubation can be accomplished under direct vision, cricothyroidotomy or tracheotomy may be necessary.
- Oxygen is given as indicated.
- The presence of shock suggests perforation and mandates an intravenous line and fluid administration.
- Damage due to alkaline corrosives occurs by liquefaction necrosis whereby the saponification of fats and solubilisation of proteins allow deep penetration into the tissue. Alkalis continue to cause damage after exposure

INGESTION:

Milk and water are the preferred diluents

No more than 2 glasses of water should be given to an adult.

- Neutralising agents should never be given since exothermic heat reaction may compound injury.
- Catharsis and emesis are absolutely contra-indicated.

* Activated charcoal does not absorb alkali.

* Gastric lavage should not be used.

Supportive care involves the following: Withhold oral feedings initially.

- If endoscopy confirms transmucosal injury start steroids only within the first 48 hours.
- Carefully evaluate the amount of tissue necrosis before assessing the need for surgical intervention.

Patients should be instructed to seek medical attention whenever they develop difficulty in swallowing (dysphagia).

SKIN AND EYE: Injury should be irrigated for 20-30 minutes.

Eye injuries require saline. [Ellenhorn & Barceloux: Medical Toxicology]

SECTION 5 Firefighting measures

Extinguishing media

- Water spray or fog.
- Foam
- Drv chemical powder.
- BCF (where regulations permit).

Fire Incompatibility

Carbon dioxide.

Special hazards arising from the substrate or mixture

None known.

Advice for firefighters	
Fire Fighting	 Alert Fire Brigade and tell them location and nature of hazard. Wear full body protective clothing with breathing apparatus. Prevent, by any means available, spillage from entering drains or water course. Use fire fighting procedures suitable for surrounding area. Do not approach containers suspected to be hot. Cool fire exposed containers with water spray from a protected location. If safe to do so, remove containers from path of fire. Equipment should be thoroughly decontaminated after use.
Fire/Explosion Hazard	 Non combustible. Not considered a significant fire risk, however containers may burn. May emit corrosive fumes.
HAZCHEM	2X

SECTION 6 Accidental release measures

Personal precautions, protective equipment and emergency procedures See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Minor Spills	 Drains for storage or use areas should have retention basins for pH adjustments and dilution of spills before discharge or disposal of material. Check regularly for spills and leaks. Remove all ignition sources. Clean up all spills immediately. Avoid contact with skin and eyes. Control personal contact with the substance, by using protective equipment. Use dry clean up procedures and avoid generating dust. Place in a suitable, labelled container for waste disposal. 		
Major Spills	 Clear area of personnel and move upwind. Alert Fire Brigade and tell them location and nature of hazard. Wear full body protective clothing with breathing apparatus. Prevent, by any means available, spillage from entering drains or water course. Consider evacuation (or protect in place). Stop leak if safe to do so. 		

Contain spill with sand, earth or vermiculite.
 Collect recoverable product into labelled containers for recycling.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 Handling and storage

Precautions for safe handling	
Safe handling	 Avoid all personal contact, including inhalation. Wear protective clothing when risk of exposure occurs. Use in a well-ventilated area. Avoid contact with moisture. Avoid contact with incompatible materials. When handling, DO NOT eat, drink or smoke. Keep containers securely sealed when not in use. Avoid physical damage to containers.
Other information	 Store in original containers. Keep containers securely sealed. Store in a cool, dry, well-ventilated area. Store away from incompatible materials and foodstuff containers. Protect containers against physical damage and check regularly for leaks. Observe manufacturer's storage and handling recommendations contained within this SDS. DO NOT store near acids, or oxidising agents No smoking, naked lights, heat or ignition sources.

Conditions for safe storage, including any incompatibilities

	DO NOT use aluminium or galvanised containers
Suitable container	 bornor use attaining of gatarised containers Lined metal can, lined metal pail/ can. Plastic pail. Polyliner drum. Packing as recommended by manufacturer. Check all containers are clearly labelled and free from leaks. For low viscosity materials Drums and jerricans must be of the non-removable head type. Where a can is to be used as an inner package, the can must have a screwed enclosure. For materials with a viscosity of at least 2680 cSt. (23 deg. C) and solids (between 15 C deg. and 40 deg C.): Removable head packaging; Cans with friction closures and low pressure tubes and cartridges may be used. Where combination packages are used, and the inner packages are of glass, porcelain or stoneware, there must be sufficient inert cushioning material in contact with inner and outer packages unless the outer packaging is a close fitting moulded plastic box and the substances are not incompatible with the plastic.
Storage incompatibility	 Calcium hydroxide produces explosive decomposition on contact with maleic anhydride may form explosive compounds or explode on contact with ammonium salts, phosphorus, nitroethane, nitromethane, nitroparaffins or nitropropane; salts may be shock-sensitive is incompatible with acids attacks some metals and coatings forms salts with nitroparaffins in the presence of water which are explosive when dried. Reacts with aluminium / zinc producing flammable, explosive hydrogen gas Avoid strong acids, acid chlorides, acid anhydrides and chloroformates. Avoid contact with copper, aluminium and their alloys.

SECTION 8 Exposure controls / personal protection

Control parameters

Occupational Exposure Limits (OEL)

INGREDIENT DATA						
Source	Ingredient	Material name	TWA	STEL	Peak	Notes
Australia Exposure Standards	calcium hydroxide Calcium hydroxide 5 mg/m3			Not Available Not Available Not Available		
Ingredient	Original IDLH			Revised IDLH		
calcium hydroxide	Not Available		Not Available			

Exposure controls

Appropriate engineering controls	Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is done to reduce the risk. Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly. The design of a ventilation system must match the particular process and chemical or contaminant in use. Employers may need to use multiple types of controls to prevent employee overexposure.
	Local exhaust ventilation is required where solids are handled as powders or crystals; even when particulates are relatively large, a certain proportion will be powdered by mutual friction.

Individual protection measures, such as personal protective equipment	
Eye and face protection	 Chemical goggles. Full face shield may be required for supplementary but never for primary protection of eyes. Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly.
Skin protection	See Hand protection below
Hands/feet protection	 Wear chemical protective gloves, e.g. PVC. Wear safety footwear or safety gumboots, e.g. Rubber The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application. The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice. Personal hygiene is a key element of effective hand care. Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended. Suitability and durability of glove type is dependent on usage.
Body protection	See Other protection below
Other protection	 Overalls. PVC Apron. PVC protective suit may be required if exposure severe. Eyewash unit. Ensure there is ready access to a safety shower.

Recommended material(s)

GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the: "Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the *computer-generated* selection:

CALCIUM HYDROXIDE

Material	CPI
NATURAL RUBBER	A
NATURAL+NEOPRENE	А

* CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

 $\ensuremath{\text{NOTE}}$: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

Ansell Glove Selection

Glove — In order of recommendation
AlphaTec® 15-554
AlphaTec® Solvex® 37-185
AlphaTec® 38-612
AlphaTec® 58-008
AlphaTec® 58-530B
AlphaTec® 58-530W
AlphaTec® 58-735
AlphaTec® 79-700
AlphaTec® Solvex® 37-675
DermaShield™ 73-711

The suggested gloves for use should be confirmed with the glove supplier.

SECTION 9 Physical and chemical properties

Information on basic physical and chemical properties

Appearance	White or off white amorphous odourless powder with bitter, alkaline taste; slightly soluble in water and insoluble in alcohol. Readily absorbs carbon dioxide from the air to form calcium carbonate; and loses water when heated strongly to form calcium oxide. Soluble in glycerol, sugar or ammonium chloride solutions. Soluble in acids with evolution of heat. Bulk density: 400-500 kg/m3. Grades available: Builders Lime, technical, Pure, BP sterilised.		
Physical state	Divided Solid	Relative density (Water = 1)	2.2-2.3

Respiratory protection

Respirators may be necessary when engineering and administrative controls do not adequately prevent exposures.

• The decision to use respiratory protection should be based on professional judgment that takes into account toxicity information, exposure measurement data, and frequency and likelihood of the worker's exposure - ensure users are not subject to high thermal loads which may result in heat stress or distress due to personal protective equipment (powered, positive flow, full face apparatus may be an option). • Published occupational exposure limits, where they exist, will assist in determining the adequacy of the selected respiratory protection. These may be government mandated or vendor recommended.

 Certified respirators will be useful for protecting workers from inhalation of particulates when properly selected and fit tested as part of a complete respiratory protection program.

• Where protection from nuisance levels of dusts are desired, use type N95 (US) or type P1 (EN143) dust masks. Use respirators and components tested and approved under appropriate government standards such as NIOSH (US) or CEN (EU)

Use approved positive flow mask if significant quantities of dust becomes airborne.
 Try to avoid creating dust conditions.

Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Applicable
pH (as supplied)	Not Applicable	Decomposition temperature (°C)	580
Melting point / freezing point (°C)	580 (-H2O)	Viscosity (cSt)	Not Applicable
Initial boiling point and boiling range (°C)	Decomposes.	Molecular weight (g/mol)	74.10
Flash point (°C)	Not Applicable	Taste	Not Available
Evaporation rate	Not Applicable	Explosive properties	Not Available
Flammability	Not Applicable	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Applicable	Surface Tension (dyn/cm or mN/m)	Not Applicable
Lower Explosive Limit (%)	Not Applicable	Volatile Component (%vol)	Negligible
Vapour pressure (kPa)	Negligible	Gas group	Not Available
Solubility in water	Partly miscible	pH as a solution (1%)	12.4 (saturated)
Vapour density (Air = 1)	Not Applicable	VOC g/L	Not Available
Heat of Combustion (kJ/g)	Not Available	Ignition Distance (cm)	Not Available
Flame Height (cm)	Not Available	Flame Duration (s)	Not Available
Enclosed Space Ignition Time Equivalent (s/m3)	Not Available	Enclosed Space Ignition Deflagration Density (g/m3)	Not Available

SECTION 10 Stability and reactivity

Reactivity	See section 7
Chemical stability	 Unstable in the presence of incompatible materials. Product is considered stable. Hazardous polymerisation will not occur.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 Toxicological information

Information on toxicological effects

Inhaled burpe Inhaled if Ingestion Th Skin Contact Fol Skin Contact Re Skin Contact Skin	linor exposures or slow dissolving of calcium hydroxide, in body fluids urning sensation. Severe acute dust inhalation may produce throat infl ersons with impaired respiratory function, airway diseases and conditi isability if excessive concentrations of particulate are inhaled. prior damage to the circulatory or nervous systems has occurred or if onducted on individuals who may be exposed to further risk if handling ihaling corrosive bases may irritate the respiratory tract. Symptoms include the material can produce chemical burns within the oral cavity and gas he material has NOT been classified by EC Directives or other classified if corroborating animal or human evidence. The material can produce chemical burns following direct contact with th kin contact is not thought to have harmful health effects (as classified illowing entry through wounds, lesions or abrasions. In the presence of moisture calcium hydroxide (slaked lime) is a caustic usult in severe burns and blistering, depending on duration of contact. eactions may not occur on exposure but response may be delayed wi pen cuts, abraded or irritated skin should not be exposed to this mate olution of material in moisture on the skin, or perspiration, may marked ntry into the blood-stream, through, for example, cuts, abrasions or les is nor to the use of the material and ensure that any external damage	ammation and fluid in the lungs. ons such as emphysema or chronic bronchitis, may incur further kidney damage has been sustained, proper screenings should be and use of the material result in excessive exposures. clude cough, choking, pain and damage to the mucous membrane. trointestinal tract following ingestion. cation systems as "harmful by ingestion". This is because of the lack he skin. under EC Directives); the material may still produce health damage irritant and can be damaging to human tissue. Skin contact may th symptoms only appearing many hours later rial ly increase skin corrosion and accelerate tissue destruction sions, may produce systemic injury with harmful effects. Examine the e is suitably protected.	
Ingestion Tr of Skin Contact Re OS Str Skin Contact	he material has NOT been classified by EC Directives or other classifier f corroborating animal or human evidence. The material can produce chemical burns following direct contact with the kin contact is not thought to have harmful health effects (as classified allowing entry through wounds, lesions or abrasions. It he presence of moisture calcium hydroxide (slaked lime) is a caustic asult in severe burns and blistering, depending on duration of contact. eactions may not occur on exposure but response may be delayed wi pen cuts, abraded or irritated skin should not be exposed to this mate olution of material in moisture on the skin, or perspiration, may marker ntry into the blood-stream, through, for example, cuts, abrasions or les in prior to the use of the material and ensure that any external damage	cation systems as "harmful by ingestion". This is because of the lack ne skin. under EC Directives); the material may still produce health damage irritant and can be damaging to human tissue. Skin contact may th symptoms only appearing many hours later rial ly increase skin corrosion and accelerate tissue destruction sions, may produce systemic injury with harmful effects. Examine the e is suitably protected.	
Skin Contact Rin Contact Skin Contact Sc Er sk	kin contact is not thought to have harmful health effects (as classified illowing entry through wounds, lesions or abrasions. It he presence of moisture calcium hydroxide (slaked lime) is a caustic esult in severe burns and blistering, depending on duration of contact. eactions may not occur on exposure but response may be delayed wi pen cuts, abraded or irritated skin should not be exposed to this mate olution of material in moisture on the skin, or perspiration, may marked ntry into the blood-stream, through, for example, cuts, abrasions or les kin prior to the use of the material and ensure that any external damage	under EC Directives); the material may still produce health damage irritant and can be damaging to human tissue. Skin contact may th symptoms only appearing many hours later rial dly increase skin corrosion and accelerate tissue destruction sions, may produce systemic injury with harmful effects. Examine the e is suitably protected.	
Th			
Eye If a	The material can produce chemical burns to the eye following direct contact. Vapours or mists may be extremely irritating. If applied to the eyes, this material causes severe eye damage. Eye contact with calcium hydroxide may result in severe irritation and pain. The material may induce ulcerations of the eyeball surface.		
Chronic Lcc Lcc	Repeated or prolonged exposure to corrosives may result in the erosion of teeth, inflammatory and ulcerative changes in the mouth and necrosis (rarely) of the jaw. Bronchial irritation, with cough, and frequent attacks of bronchial pneumonia may ensue. Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure. Long term exposure to calcium hydroxide may result in narrowing of the gullet, with difficulty in swallowing. This may happen after weeks, months or years of exposure. Long term exposure to high dust concentrations may cause changes in lung function i.e. pneumoconiosis, caused by particles less than 0.5 micron penetrating and remaining in the lung.		
CALCIUM HYDROXIDE	τοχιςιτγ	IRRITATION	
	dermal (rat) LD50: >2000 mg/kg ^[1]	Eye (Rodent - rabbit): 10mg - Severe	
	Inhalation (Rat) LC50: >3 mg/l4h ^[1]	Eye: adverse effect observed (irritating) ^[1]	

	Oral (Rat) LD50: >2000 mg/kg ^[1]	Skin: adverse eff	ect observed (irritating) ^[1]
Legend:	1. Value obtained from Europe ECHA Registered Su specified data extracted from RTECS - Register of To		ained from manufacturer's SDS. Unless other
	The material may produce severe irritation to the eye produce conjunctivitis. Asthma-like symptoms may continue for months or e condition known as reactive airways dysfunction sym compound. Main criteria for diagnosing RADS include	ven years after exposure to the mate drome (RADS) which can occur after e the absence of previous airways dis	rial ends. This may be due to a non-allergic exposure to high levels of highly irritating sease in a non-atopic individual, with sudden o
CALCIUM HYDROXIDE	of persistent asthma-like symptoms within minutes to include a reversible airflow pattern on lung function te and the lack of minimal lymphocytic inflammation, wit disorder with rates related to the concentration of and is a disorder that occurs as a result of exposure due reversible after exposure ceases. The disorder is char-	ests, moderate to severe bronchial hy thout eosinophilia. RADS (or asthma) d duration of exposure to the irritating to high concentrations of irritating sub	perreactivity on methacholine challenge testin following an irritating inhalation is an infreque substance. On the other hand, industrial brom ostance (often particles) and is completely
CALCIUM HYDROXIDE	include a reversible airflow pattern on lung function te and the lack of minimal lymphocytic inflammation, wild disorder with rates related to the concentration of and is a disorder that occurs as a result of exposure due	ests, moderate to severe bronchial hy thout eosinophilia. RADS (or asthma) d duration of exposure to the irritating to high concentrations of irritating sub	perreactivity on methacholine challenge testin following an irritating inhalation is an infreque substance. On the other hand, industrial bron ostance (often particles) and is completely
	include a reversible airflow pattern on lung function te and the lack of minimal lymphocytic inflammation, wit disorder with rates related to the concentration of and is a disorder that occurs as a result of exposure due reversible after exposure ceases. The disorder is char-	ests, moderate to severe bronchial hy thout eosinophilia. RADS (or asthma) d duration of exposure to the irritating to high concentrations of irritating sub aracterized by difficulty breathing, cou	rperreactivity on methacholine challenge testin following an irritating inhalation is an infreque substance. On the other hand, industrial bron ostance (often particles) and is completely gh and mucus production.
Acute Toxicity	include a reversible airflow pattern on lung function te and the lack of minimal lymphocytic inflammation, wit disorder with rates related to the concentration of and is a disorder that occurs as a result of exposure due i reversible after exposure ceases. The disorder is cha	ests, moderate to severe bronchial hy thout eosinophilia. RADS (or asthma) d duration of exposure to the irritating to high concentrations of irritating sub aracterized by difficulty breathing, cou Carcinogenicity	rperreactivity on methacholine challenge testin following an irritating inhalation is an infreque substance. On the other hand, industrial bron sstance (often particles) and is completely igh and mucus production.
Acute Toxicity Skin Irritation/Corrosion Serious Eye	include a reversible airflow pattern on lung function te and the lack of minimal lymphocytic inflammation, wit disorder with rates related to the concentration of and is a disorder that occurs as a result of exposure due to reversible after exposure ceases. The disorder is char	ests, moderate to severe bronchial hy thout eosinophilia. RADS (or asthma) d duration of exposure to the irritating to high concentrations of irritating sub aracterized by difficulty breathing, cour Carcinogenicity Reproductivity	rperreactivity on methacholine challenge testin following an irritating inhalation is an infreque substance. On the other hand, industrial bron ostance (often particles) and is completely gh and mucus production.

Data available to make classification

SECTION 12 Ecological information

Toxicity

	Endpoint	Test Duration (hr)	Species	Value	Source
	EC50	72h	Algae or other aquatic plants	>14mg/l	2
CALCIUM HYDROXIDE	EC50	48h	Crustacea	49.1mg/l	2
	NOEC(ECx)	72h	Algae or other aquatic plants	14mg/l	2
	LC50	96h	Fish	33.884mg/L	4
Legend:	Ecotox database		rred Substances - Ecotoxicological Information - zard Assessment Data 6. NITE (Japan) - Biocon		

Prevent, by any means available, spillage from entering drains or water courses. DO NOT discharge into sewer or waterways.

Persistence and degradability

· · · · · · · · · · · · · · · · · · ·		
Ingredient	Persistence: Water/Soil	Persistence: Air
	No Data available for all ingredients	No Data available for all ingredients
Bioaccumulative potential		
Ingredient	Bioaccumulation	
	No Data available for all ingredients	
Mobility in soil		
Ingredient	Mobility	
	No Data available for all ingredients	

SECTION 13 Disposal considerations

Product / Packaging disposal	Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked. A Hierarchy of Controls seems to be common - the user should investigate: • Reduction • Reuse • Recycling • Disposal (if all else fails) This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate. In most instances the supplier of the material should be consulted. • DO NOT allow wash water from cleaning or process equipment to enter drains. • It may be necessary to collect all wash water for treatment before disposal. • In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first. • Where in doubt contact the responsible authority.
	 Recycle wherever possible. Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified. Treat and neutralise at an approved treatment plant.

Treatment should involve: Mixing or slurrying in water; Neutralisation with suitable dilute acid followed by: burial in a land-fill specifically licensed to accept chemical and / or pharmaceutical wastes or Incineration in a licensed apparatus (after admixture with suitable combustible material).
Decontaminate empty containers. Observe all label safeguards until containers are cleaned and destroyed.

SECTION 14 Transport information

Labels Required



HAZCHEM 2X

Marine Pollutant

Land transport (ADG)

14.1. UN number or ID number	3262		
14.2. UN proper shipping name	CORROSIVE SOLID, BASIC, INORGANIC, N.O.S. (contains calcium hydroxide)		
14.3. Transport hazard class(es)	Class Subsidiary Hazard	8 Not Applicable	
14.4. Packing group	II		
14.5. Environmental hazard	Not Applicable		
14.6. Special precautions for user	Special provisions Limited quantity	223 274 5 kg	

Air transport (ICAO-IATA / DGR)

14.1. UN number	3262			
14.2. UN proper shipping name	Corrosive solid, basic, inorganic, n.o.s. * (contains calcium hydroxide)			
	ICAO/IATA Class	Class 8		
14.3. Transport hazard class(es)	ICAO / IATA Subsidiary Hazard	Not Applicable		
01033(03)	ERG Code	8L		
14.4. Packing group	Ш			
14.5. Environmental hazard	Not Applicable			
	Special provisions		A3 A803	
	Cargo Only Packing Instructions		864	
14.6. Special precautions for user	Cargo Only Maximum Qty / Pack		100 kg	
	Passenger and Cargo Packing Instructions		860	
	Passenger and Cargo Maximum Qty / Pack		25 kg	
	Passenger and Cargo Limited Quantity Packing Instructions		Y845	
	Passenger and Cargo Limited Maximum Qty / Pack		5 kg	

Sea transport (IMDG-Code / GGVSee)

14.1. UN number	3262		
14.2. UN proper shipping name	CORROSIVE SOLID, BASIC, INORGANIC, N.O.S. (contains calcium hydroxide)		
14.3. Transport hazard class(es)	IMDG Class IMDG Subsidiary Haz	8 eard Not Applicable	
14.4. Packing group	Ш		
14.5 Environmental hazard	Not Applicable		
14.6. Special precautions for user	EMS Number Special provisions Limited Quantities	F-A , S-B 223 274 5 kg	

14.7.1. Transport in bulk according to Annex II of MARPOL and the IBC code

Product name	Pollution Category	Ship Type
Calcium hydroxide slurry	Υ	2

14.7.2. Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
calcium hydroxide	Not Available
4.7.3. Transport in bulk in	accordance with the IGC Code
4.7.3. Transport in bulk in Product name	accordance with the IGC Code Ship Type

SECTION 15 Regulatory information

Safety, health and environmental regulations / legislation specific for the substance or mixture

calcium hydroxide is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals

Australian Inventory of Industrial Chemicals (AIIC)

International WHO List of Proposed Occupational Exposure Limit (OEL) Values for Manufactured Nanomaterials (MNMS)

Additional Regulatory Information

Not Applicable

National Inventory Status

National Inventory	Status	
Australia - AIIC / Australia Non- Industrial Use	Yes	
Canada - DSL	Yes	
Canada - NDSL	No (calcium hydroxide)	
China - IECSC	Yes	
Europe - EINEC / ELINCS / NLP	Yes	
Japan - ENCS	Yes	
Korea - KECI	Yes	
New Zealand - NZIoC	Yes	
Philippines - PICCS	Yes	
USA - TSCA	All chemical substances in this product have been designated as TSCA Inventory 'Active'	
Taiwan - TCSI	Yes	
Mexico - INSQ	Yes	
Vietnam - NCI	Yes	
Russia - FBEPH	Yes	
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.	

SECTION 16 Other information

Revision Date	20/06/2022
Initial Date	16/08/2006
SDS Version Summary	

Version	Date of Update	Sections Updated
3.1	20/06/2022	Expiration. Review and Update

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

Definitions and abbreviations

- PC TWA: Permissible Concentration-Time Weighted Average
- PC STEL: Permissible Concentration-Short Term Exposure Limit
- IARC: International Agency for Research on Cancer
- ACGIH: American Conference of Governmental Industrial Hygienists
- STEL: Short Term Exposure Limit
- TEEL: Temporary Emergency Exposure Limit.
- IDLH: Immediately Dangerous to Life or Health Concentrations
- ES: Exposure Standard
- OSF: Odour Safety Factor
- NOAEL: No Observed Adverse Effect Level
- LOAEL: Lowest Observed Adverse Effect Level
- TLV: Threshold Limit Value
- LOD: Limit Of Detection
- OTV: Odour Threshold Value
- BCF: BioConcentration Factors

- BEI: Biological Exposure Index
 DNEL: Derived No-Effect Level
- PNEC: Predicted no-effect concentration
- AIIC: Australian Inventory of Industrial Chemicals
- DSL: Domestic Substances List
 NDSL: Non-Domestic Substances List
- IECSC: Inventory of Existing Chemical Substance in China
- EINECS: European INventory of Existing Commercial chemical Substances
- ELINCS: European List of Notified Chemical Substances
- NLP: No-Longer Polymers
 ENCS: Existing and New Chemical Substances Inventory
 KECI: Korea Existing Chemicals Inventory
 NZIoC: New Zealand Inventory of Chemicals
- PICCS: Philippine Inventory of Chemicals and Chemical Substances
- TSCA: Toxic Substances Control Act
- TCSI: Taiwan Chemical Substance Inventory
 INSQ: Inventario Nacional de Sustancias Químicas
- NCI: National Chemical Inventory
- FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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