

# Hydrogen peroxide 30% (100 vol) ALPHA CHEMICALS PTY LTD

Chemwatch: **4023296** Version No: **5.1.7.7** 

Safety Data Sheet according to WHS Regulations (Hazardous Chemicals) Amendment 2020 and ADG requirements

Chemwatch Hazard Alert Code: 3

Issue Date: **30/12/2020** Print Date: **21/06/2021** S.GHS.AUS.EN

#### SECTION 1 Identification of the substance / mixture and of the company / undertaking

## Product Identifier

Product identifier			
Product name	Hydrogen peroxide 30% (100 vol)		
Chemical Name	Not Applicable		
Synonyms	Name Product Code; HYDROGEN PEROXIDE 30% 260; HYDROGEN PEROXIDE 305 100 VOL 922; HYDROGEN PEROXIDE 35% 120 VOL 923; HYDROGEN PEROXIDE 35% 2430; HYDROGEN PEROXIDE 100 VOL M&B 10328; HYDROGEN PEROXIDE 35% 120 VOL 10525; INTEROX AG BATH/HYDROGEN PEROXIDE 11004; INTEROX 50CG/HYDROGEN PEROXIDE 11005; HYDROGEN PEROXIDE 30% 100 VOL A5500; HYDROGEN PEROXIDE 30% 100 VOL BSPAS; Product Code: A525		
Proper shipping name	HYDROGEN PEROXIDE, AQUEOUS SOLUTION with not less than 20% but not more than 60% hydrogen peroxide (stabilised as necessary)		
Chemical formula	Not Applicable		
Other means of identification	Not Available		

#### Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses Bleaching agents, chemical industry, electronic industry, metal treatment, odour agents, textile industry, water treatment and pulp and paper.

#### Details of the supplier of the safety data sheet

Registered company name	ALPHA CHEMICALS PTY LTD		
Address	ALLEN PLACE WETHERILL PARK NSW 2099 Australia		
Telephone	61 (0)2 9982 4622		
Fax	Not Available		
Website	~		
Email	shane@alphachem.com.au		

#### Emergency telephone number

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Association / Organisation	ALPHA CHEMICALS PTY LTD
Emergency telephone numbers	61 (0)418 237 771
Other emergency telephone numbers	Not Available

## **SECTION 2 Hazards identification**

#### Classification of the substance or mixture

# HAZARDOUS CHEMICAL. DANGEROUS GOODS. According to the WHS Regulations and the ADG Code.

#### ChemWatch Hazard Ratings

		Min	Max	
Flammability	0			
Toxicity	2			0 = Minimum
Body Contact	3			1 = Low
Reactivity	2			2 = Moderate
Chronic	0		1	3 = High 4 = Extreme

Poisons Schedule	S6
Classification <sup>[1]</sup>	Acute Toxicity (Oral) Category 4, Skin Corrosion/Irritation Category 1A, Serious Eye Damage/Eye Irritation Category 1, Acute Toxicity (Inhalation) Category 4, Specific target organ toxicity - single exposure Category 3 (respiratory tract irritation), Acute Aquatic Hazard Category 3, Oxidizing Liquid Category 2
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI



Signal word Danger

#### Hazard statement(s)

H302	Harmful if swallowed.	
H314	Causes severe skin burns and eye damage.	
H332	Harmful if inhaled.	
H335	May cause respiratory irritation.	
H402	Harmful to aquatic life.	
H272	May intensify fire; oxidiser.	

#### Precautionary statement(s) Prevention

P210	Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking.		
P260	Do not breathe mist/vapours/spray.		
P264	Wash all exposed external body areas thoroughly after handling.		
P271	lse only outdoors or in a well-ventilated area.		
P280	Wear protective gloves, protective clothing, eye protection and face protection.		
P220	P220 Keep away from clothing and other combustible materials.		
P270	Do not eat, drink or smoke when using this product.		
P273	Avoid release to the environment.		

## Precautionary statement(s) Response

P301+P330+P331	IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.		
P303+P361+P353	F ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water [or shower].		
P305+P351+P338	F IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.		
P310	Immediately call a POISON CENTER/doctor/physician/first aider.		
P370+P378	In case of fire: Use alcohol resistant foam or fine spray/water fog to extinguish.		
P363	Wash contaminated clothing before reuse.		
P301+P312	IF SWALLOWED: Call a POISON CENTER/doctor/physician/first aider if you feel unwell.		
P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.		

#### Precautionary statement(s) Storage

P405	Store locked up.
P403+P233	Store in a well-ventilated place. Keep container tightly closed.

## Precautionary statement(s) Disposal

P501 Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

## **SECTION 3 Composition / information on ingredients**

#### Substances

See section below for composition of Mixtures

#### Mixtures

CAS No	%[weight]	Name
7722-84-1	30-60	hydrogen peroxide
Not Available	30-60	ingredients determined not to be hazardous, including
7732-18-5		Distilled Water
Legend:	1. Classified by Chemwatch; Classification drawn from C8	2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4. L: * EU IOELVs available

## **SECTION 4 First aid measures**

Description of first aid measures		
Eye Contact	<ul> <li>If this product comes in contact with the eyes:</li> <li>Immediately hold eyelids apart and flush the eye continuously with running water.</li> <li>Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.</li> <li>Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes.</li> <li>Transport to hospital or doctor without delay.</li> </ul>	

	Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	<ul> <li>If skin or hair contact occurs:</li> <li>Immediately flush body and clothes with large amounts of water, using safety shower if available.</li> <li>Quickly remove all contaminated clothing, including footwear.</li> <li>Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre.</li> <li>Transport to hospital, or doctor.</li> </ul>
Inhalation	<ul> <li>If fumes or combustion products are inhaled remove from contaminated area.</li> <li>Lay patient down. Keep warm and rested.</li> <li>Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.</li> <li>Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.</li> <li>Transport to hospital, or doctor.</li> <li>Inhalation of vapours or aerosols (mists, fumes) may cause lung oedema.</li> <li>Corrosive substances may cause lung damage (e.g. lung oedema, fluid in the lungs).</li> <li>As this reaction may be delayed up to 24 hours after exposure, affected individuals need complete rest (preferably in semi-recumbent posture) and must be kept under medical observation even if no symptoms are (yet) manifested.</li> <li>Before any such manifestation, the administration of a spray containing a dexamethasone derivative or beclomethasone derivative may be considered.</li> <li>This must definitely be left to a doctor or person authorised by him/her.</li> <li>(ICSC13719)</li> </ul>
Ingestion	<ul> <li>For advice, contact a Poisons Information Centre or a doctor at once.</li> <li>Urgent hospital treatment is likely to be needed.</li> <li>If swallowed do NOT induce vomiting.</li> <li>If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.</li> <li>Observe the patient carefully.</li> <li>Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.</li> <li>Give water to rinse out mouth, then provide liquid slowly and as much as casuality can comfortably drink.</li> <li>Transport to hospital or doctor without delay.</li> </ul>

#### Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

- Hydrogen peroxide at moderate concentrations (5% or more) is a strong oxidant.
- Direct contact with the eye is likely to cause corneal damage especially if not washed immediately. Careful ophthalmologic evaluation is recommended and the possibility of local corticosteroid therapy should be considered.
- Because of the likelihood of systemic effects attempts at evacuating the stomach via emesis induction or gastric lavage should be avoided.
- There is remote possibility, however, that a nasogastric or orogastric tube may be required for the reduction of severe distension due to gas formation"
- Fisher Scientific SDS

Depending on the degree of exposure, periodic medical examination is indicated. The symptoms of lung oedema often do not manifest until a few hours have passed and they are aggravated by physical effort. Rest and medical observation is therefore essential. Immediate administration of an appropriate spray, by a doctor or a person authorised by him/her should be considered. (ICSC24419/24421

#### **SECTION 5 Firefighting measures**

#### Extinguishing media

- FOR SMALL FIRE:
- ▶ USE FLOODING QUANTITIES OF WATER.
- DO NOT use dry chemical, CO2, foam or halogenated-type extinguishers.
- FOR LARGE FIRE
- Flood fire area with water from a protected position

#### Special hazards arising from the substrate or mixture

Fire Incompatibility	<ul> <li>Avoid storage with reducing agents.</li> <li>Avoid any contamination of this material as it is very reactive and any contamination is potentially hazardous</li> </ul>
Advice for firefighters	
	Alert Fire Brigade and tell them location and nature of hazard.

Fire Fighting	<ul> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>Wear breathing apparatus plus protective gloves in the event of a fire.</li> <li>Prevent, by any means available, spillage from entering drains or water courses.</li> <li>Use fire fighting procedures suitable for surrounding area.</li> <li>DO NOT approach containers suspected to be hot.</li> <li>Cool fire exposed containers with water spray from a protected location.</li> <li>If safe to do so, remove containers from path of fire.</li> </ul>
	Equipment should be thoroughly decontaminated after use.
Fire/Explosion Hazard	<ul> <li>Will not burn but increases intensity of fire.</li> <li>Heating may cause expansion or decomposition leading to violent rupture of containers.</li> <li>Heat affected containers remain hazardous.</li> <li>Contact with combustibles such as wood, paper, oil or finely divided metal may produce spontaneous combustion or violent decomposition.</li> <li>May emit irritating, poisonous or corrosive fumes.</li> </ul>
HAZCHEM	2P

#### **SECTION 6 Accidental release measures**

Personal precautions, protective equipment and emergency procedures See section 8

See section 12

# Methods and material for containment and cleaning up

Minor Spills	<ul> <li>Drains for storage or use areas should have retention basins for pH adjustments and dilution of spills before discharge or disposal of material.</li> <li>Check regularly for spills and leaks.</li> <li>Clean up all spills immediately.</li> <li>No smoking, naked lights, ignition sources.</li> <li>Avoid all contact with any organic matter including fuel, solvents, sawdust, paper or cloth and other incompatible materials, as ignition may result.</li> <li>Avoid breathing dust or vapours and all contact with skin and eyes.</li> <li>Control personal contact with the substance, by using protective equipment.</li> <li>Contain and absorb spill with dry sand, earth, inert material or vermiculite.</li> <li>DO NOT use sawdust as fire may result.</li> <li>Scoop up solid residues and seal in labelled drums for disposal.</li> </ul>
Major Spills	<ul> <li>Clear area of personnel and move upwind.</li> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>May be violently or explosively reactive.</li> <li>Wear full body protective clothing with breathing apparatus.</li> <li>Prevent, by any means available, spillage from entering drains or water course.</li> <li>Consider evacuation (or protect in place).</li> <li>No smoking, flames or ignition sources.</li> <li>Increase ventilation.</li> </ul>

Personal Protective Equipment advice is contained in Section 8 of the SDS.

# SECTION 7 Handling and storage

## Precautions for safe handling

Safe handling	<ul> <li>DO NOT allow clothing wet with material to stay in contact with skin</li> <li>For oxidisers, including peroxides.</li> <li>Avoid personal contact and inhalation of dust, mist or vapours.</li> <li>Provide adequate ventilation.</li> <li>Always wear protective equipment and wash off any spillage from clothing.</li> <li>Keep material away from light, heat, flammables or combustibles.</li> <li>Keep cool, dry and away from incompatible materials.</li> <li>Avoid physical damage to containers.</li> <li>DO NOT repack or return unused portions to original containers.</li> </ul>
Other information	<ul> <li>Store in original containers.</li> <li>Keep containers securely sealed as supplied.</li> <li>Store in a cool, well ventilated area.</li> <li>Keep dry.</li> <li>Store under cover and away from sunlight.</li> <li>Store away from incompatible materials, debris and waste. Contact may cause fire or violent reaction.</li> <li>Store away from incompatible materials and foodstuff containers.</li> <li>In addition, Goods of Class 5.1, packing group II should be:</li> <li>stored in piles so that</li> <li>the height of the pile does not exceed 1 metre</li> <li>the maximum quantity in a pile or building does not exceed 1000 tonnes unless the area is provided with automatic fire extinguishers</li> <li>the maximum distance between piles is not less than 2 metres where the room is provided with automatic fire extinguishers or 3 meters if not.</li> <li>the minimum distance to walls is not less than 1 metre.</li> </ul>

## Conditions for safe storage, including any incompatibilities

Suitable container	<ul> <li>DO NOT repack. Use containers supplied by manufacturer only.</li> <li>For low viscosity materials</li> <li>Drums and jerricans must be of the non-removable head type.</li> <li>Where a can is to be used as an inner package, the can must have a screwed enclosure.</li> <li>For materials with a viscosity of at least 2680 cSt. (23 deg. C) and solids: <ul> <li>Removable head packaging and</li> <li>cans with friction closures may be used.</li> </ul> </li> <li>Where combination packages are used, and the inner packages are of glass, there must be sufficient inert cushioning material in contact with inner and outer packages *.</li> <li>In addition, where inner packagings are glass and contain liquids of packing group I and II there must be sufficient inert absorbent to absorb any spillage *.</li> <li>* unless the outer packaging is a close fitting moulded plastic box and the substances are not incompatible with the plastic.</li> <li>Hydrogen peroxide containing/ generating materials requiring rigid packaging.</li> <li>Store in:</li> <li>* containers with vented lids.</li> <li>* properly passivated atuminium containers.</li> <li>* properly passivated stainless steel.</li> <li>* polyethylene containers.</li> <li>* procelain, vitreous stoneware</li> <li>* Teflon lined containers.</li> </ul>
Storage incompatibility	<ul> <li>Hydrogen peroxide</li> <li>is a powerful oxidiser</li> <li>contamination or heat may cause self accelerating exothermic decomposition with oxygen gas and steam release - this may generate dangerous pressures - steam explosion.</li> <li>reacts dangerously with rust, dust, dirt, iron, copper, acids, metals and salts, organic material.</li> <li>is unstable if heated. (e.g): one volume of 70% hydrogen peroxide solution decomposes to produce 300 volumes of oxygen gas.</li> </ul>

ECTION 8 Exposure contro	<ul> <li>concentrated or pur contaminated; or if</li> <li>mixtures with comb towels or mop may</li> <li>reacts violently with powder, permangar hydrogen selenide, hydrazine, nickel m quinoline, tetrahydrr combustible materia</li> <li>reacts with cobalt, or and other catalytic re violent catalytic dec manganese or their</li> <li>forms unstable and diphenyl diselenide</li> <li>Avoid strong bases</li> <li>Avoid storage with re</li> </ul>	reducing agents, alcoho nates, acetone, benzene iron oxides, lithium tetra onoxide, nitrogenous ba othiophene, tin(II) chlorid als; avoid contact with or oopper and its alloys, chr netals, metal oxides and omposition will occur in salts. possible explosive mate , ethyl acetate, glycols, k ation of this material as reducing agents.	heat and decompose s ivironment, especially in sult in spontaneous com ols, ammonia, carboxylii sulfonic anhydride, 1,1- hydroaluminate, magne ses, osmium tetraoxide de, thiodiglycol, thiophar imbustibles including lu omium, iridium, iron, lea d salts - avoid metallic b contact with certain met erials with acetic anhydri tetene, ketones, triethyll	pontaneou n the presen- hustion or c acids, ac dimethylhy sisium tetra a lapha-phi- he, tin(II) c bricants ar ad, manga owls and s als such a ide, aconit tin hydrope	ence of metal ions r may be impact- or h etic acid, cobalt oxidi ydrazine, dimethylphe hydroaluminate, man enylselenoketones, p shloride, unsaturated ind graphite inese, Monel, osmium stirrers. Is iron, copper, chrom ic acid, aniline, carbo eroxide, 1,3,5-trioxan	eat- sensitive - evapo es, copper(II) chloride enylphosphine, gadoli ganese(II) oxide, me hosphorus, phosphor organic compounds, i n, palladium, platinum ium, brass, bronze, li xylic acids, 1,4-diaza e, vinyl acetate.	e, ethers, metal nium hydroxide, rcury oxide, methyl us(V) oxide, readily oxidisable and , gold, silver, zinc, ead, silver,
	p p						
ontrol parameters							
Occupational Exposure Limits (C	DEL)						
INGREDIENT DATA							
Source	Ingredient	Material name	TWA		STEL	Peak	Notes
Australia Exposure Standards	hydrogen peroxide	Hydrogen peroxide	1 ppm / 1.4 mg	j/m3	Not Available	Not Available	Not Available
Emergency Limits							
Ingredient	TEEL-1		TEEL-2		TE	EL-3	
hydrogen peroxide	Not Available		Not Available		No	ot Available	
Ingredient	Original IDLH			Revis	ed IDLH		
hydrogen peroxide	75 ppm				vailable		
Distilled Water	Not Available			Not Available			
Appropriate engineering controls	Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is done to reduce the risk. Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly. The design of a ventilation system must match the particular process and chemical or contaminant in use. Employers may need to use multiple types of controls to prevent employee overexposure.						
Personal protection	General exhaust is adequate under normal operating conditions.						
Eye and face protection	<ul> <li>Chemical goggles.</li> <li>Full face shield may be required for supplementary but never for primary protection of eyes.</li> <li>Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly.</li> </ul>						
Skin protection	See Hand protection be	•		-			
Hands/feet protection	<ul> <li>Wear chemical protective gloves, e.g. PVC.</li> <li>Wear safety footwear or safety gunboots, e.g. Rubber</li> <li>When handling corrosive liquids, wear trousers or overalls outside of boots, to avoid spills entering boots.</li> <li>The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application.</li> <li>The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice.</li> <li>Personal hygiene is a key element of effective hand care. Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended.</li> <li>Suitability and durability of glove type is dependent on usage.</li> <li>Leather wear not recommended: Contaminated leather footwear, watch bands, should be destroyed, i.e. burnt, as they cannot be adequately decontaminated</li> <li>Where hydrogen peroxide exposure may occur do NOT wear PVA gloves.</li> <li>DO NOT use leather or cotton gloves, leather shoes as spill may cause fire.</li> <li>Care: Effects may be delayed.</li> </ul>						

Body protection	See Other protection below
Other protection	<ul> <li>Overalls.</li> <li>PVC Apron.</li> <li>PVC protective suit may be required if exposure severe.</li> <li>Eyewash unit.</li> <li>Ensure there is ready access to a safety shower.</li> <li>Some plastic personal protective equipment (PPE) (e.g. gloves, aprons, overshoes) are not recommended as they may produce static electricity.</li> <li>For large scale or continuous use wear tight-weave non-static clothing (no metallic fasteners, cuffs or pockets).</li> <li>Non sparking safety or conductive footwear should be considered. Conductive footwear describes a boot or shoe with a sole made from a conductive compound chemically bound to the bottom components, for permanent control to electrically ground the foot an shall dissipate static electricity from the body to reduce the possibility of ignition of volatile compounds. Electrical resistance must range between 0 to 500,000 ohms. Conductive shoes should be stored in lockers close to the room in which they are worn. Personnel who have been issued conductive footwear should not wear them from their place of work to their homes and return.</li> </ul>

#### Recommended material(s)

#### GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the: "Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the *computer-generated* selection:

Hydrogen peroxide 30% (100 vol)

Material	CPI
NAT+NEOPR+NITRILE	A
NATURAL RUBBER	A
NATURAL+NEOPRENE	A
NEOPRENE	A
NEOPRENE/NATURAL	A
NITRILE	A
PVC	A

\* CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

NOTE: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

\* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

## **SECTION 9 Physical and chemical properties**

#### Information on basic physical and chemical properties

Appearance Clear colourless liquid with a pungent odour; mixes with water. Soluble in polar organic solvents

Appearance	Clear colouriess liquid with a pungent odour, mixes with water. Soluble in polar organic solvents.				
Physical state	Liquid	Relative density (Water = 1)	1.2 (water=50%)		
Odour	Not Available	Partition coefficient n-octanol / water	Not Available		
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Applicable		
pH (as supplied)	1-4	Decomposition temperature	Not Available		
Melting point / freezing point (°C)	-52 (freezing pt)	Viscosity (cSt)	1.17 @20C		
Initial boiling point and boiling range (°C)	115	Molecular weight (g/mol)	Not Applicable		
Flash point (°C)	Not Applicable	Taste	Not Available		
Evaporation rate	Not Available	Explosive properties	Not Available		
Flammability	Not Applicable	Oxidising properties	Not Available		
Upper Explosive Limit (%)	Not Applicable	Surface Tension (dyn/cm or mN/m)	Not Available		
Lower Explosive Limit (%)	Not Applicable	Volatile Component (%vol)	Not Available		
Vapour pressure (kPa)	Negligible	Gas group	Not Available		
Solubility in water	Miscible	pH as a solution (%)	Not Available		
Vapour density (Air = 1)	1	VOC g/L	Not Available		

#### **SECTION 10 Stability and reactivity**

Reactivity See section 7

# Respiratory protection

Type B Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required. Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	B-AUS	-	B-PAPR-AUS / Class 1
up to 50 x ES	-	B-AUS / Class 1	-
up to 100 x ES	-	B-2	B-PAPR-2 ^

#### ^ - Full-face

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable under normal handling conditions.</li> <li>Prolonged exposure to heat.</li> <li>Hazardous polymerisation will not occur.</li> <li>Solutions of hydrogen peroxide slowly decompose, releasing oxygen, and so are often stabilised by the addition of acetanilide, etc.</li> </ul>
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

#### **SECTION 11 Toxicological information**

#### Information on toxicological effects

Inhaled	Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may be harmful. Inhalation of quantities of liquid mist may be extremely hazardous, even lethal due to spasm, extreme irritation of larynx and bronchi, chemical pneumonitis and pulmonary oedema. Inhaling excessive levels of mist may result in headache, dizziness, vomiting, diarrhoea, irritability, sleeplessness and fluid in the lungs, and cause extreme irritation of the nose and chest, cough, discomfort, shortness of breath and inflammation of the nose and throat. Whole-body effects of hydrogen peroxide poisoning include tremor, numbness of the limbs, convulsions, coma and shock. Hydrogen peroxide has poor warning properties. Corrosive acids can cause irritation of the respiratory tract, with coughing, choking and mucous membrane damage. There may be dizziness, headache, nausea and weakness.
Ingestion	Accidental ingestion of the material may be harmful; animal experiments indicate that ingestion of less than 150 gram may be fatal or may produce serious damage to the health of the individual. The material can produce chemical burns within the oral cavity and gastrointestinal tract following ingestion. Hydrogen peroxide may cause blistering and bleeding from the throat and stomach. When swallowed, it may release large quantities of oxygen which could hyper-distend the stomach and gut and may cause internal bleeding, mouth and throat burns and rupture of the gut. There may also be fever, nausea, foaming at the mouth, voriting, chest and stomach pain, loss of consciousness, and movement disorders and death. Large amounts can also cause cessation of breath, dizziness, headache, tremors weakness or numbness in the extremities and convulsions. Hydrogen peroxide concentrate is corrosive and must not be taken undiluted.
Skin Contact	Skin contact with the material may be harmful; systemic effects may result following absorption. The material can produce chemical burns following direct contact with the skin. Skin contact will result in rapid drying, bleaching, leading to chemical burns on prolonged contact Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected. Hydrogen peroxide is used topically as dental gel and to clean minor wounds. It may cause dose dependent effect on the skin including bleaching, blistering, reddening and corrosion ( at >50% concentration).
Eye	The material can produce chemical burns to the eye following direct contact. Vapours or mists may be extremely irritating. If applied to the eyes, this material causes severe eye damage. Hydrogen peroxide concentrations above 10% are corrosive to the eye and may cause corneal ulceration even days after exposure.
Chronic	There has been some concern that this material can cause cancer or mutations but there is not enough data to make an assessment. Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure. Hydrogen peroxide as a human food additive is generally regarded as safe, when used with certain limitations. In experimental animals hydrogen peroxide given by mouth causes damage to the teeth, liver, kidney, stomach and bowel. Inhalation exposure to hydrogen peroxide caused skin irritation, sneezing and death in animals. Skin irritation, sneezing, excessive secretion of tears, and whitening of the hair was also seen in animals chronically exposed to hydrogen peroxide. Repeated or prolonged exposure to acids may result in the erosion of teeth, swelling and/or ulceration of mouth lining. Irritation of airways to lung, with cough, and inflammation of lung tissue often occurs.

	TOXICITY	IRRITATION	
Hydrogen peroxide 30% (100 vol)	Dermal (Rabbit) LD50: 2000 mg/kg <sup>[2]</sup>	Not Available	
V01)	Oral (Rat) LD50: 1232 mg/kg (35% water) <sup>[2]</sup>		
	ΤΟΧΙΟΙΤΥ	IRRITATION	
	Dermal (rabbit) LD50: >2000 mg/kg <sup>[1]</sup>	Not Available	
hydrogen peroxide	Inhalation(Rat) LC50; 2 mg/L4h <sup>[2]</sup>		
	Oral(Rat) LD50; >225 mg/kg <sup>[2]</sup>		
	ΤΟΧΙΟΙΤΥ	IRRITATION	
Distilled Water	Oral(Rat) LD50; >90000 mg/kg <sup>[2]</sup>	Not Available	
Legend:	<ol> <li>Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances</li> </ol>		

Asthma-like symptoms may continue for months or even years after exposure to the material ends. This may be due to a non-allergic condition known as reactive airways dysfunction syndrome (RADS) which can occur after exposure to high levels of highly irritating compound. Main criteria for diagnosing RADS include the absence of previous airways disease in a non-atopic individual, with sudden onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. Other criteria for diagnosis of RADS include a reversible airflow pattern on lung function tests, moderate to severe bronchial hyperreactivity on methacholine challenge testing, and the lack of minimal lymphocytic inflammation, without eosinophilia. RADS (or asthma) following an irritating inhalation is an infrequent disorder with rates related to the concentration of and duration of exposure to the irritating substance. On the other hand, industrial bronchitis is a disorder that occurs as a result of exposure due to high concentrations of irritating substance (often particles) and is completely reversible after exposure ceases. The

Continued...

## Hydrogen peroxide 30% (100 vol)

	disorder is characterized by difficulty breathing, cough and mucus production. Exposure to hydrogen peroxide via the skin or oral route can produce toxic effects. Animal studies have shown evidence of damage to the kidney, gut, thymus and liver. Stomach and intestinal lesions including benign and malignant cancers have been observed in mice. It may produce genetic and developmental defects but no reproductive toxicity was reported in mice. The substance is classified by IARC as Group 3: <b>NOT</b> classifiable as to its carcinogenicity to humans. Evidence of carcinogenicity may be inadequate or limited in animal testing.				
HYDROGEN PEROXIDE & DISTILLED WATER	No significant acute toxicological data identified in literature search.				
Acute Toxicity	<b>*</b>	Carcinogenicity	×		
Skin Irritation/Corrosion	×	Reproductivity	×		
Serious Eye Damage/Irritation	×	STOT - Single Exposure	✓		
Respiratory or Skin sensitisation	×	STOT - Repeated Exposure	×		
Mutagenicity	×	Aspiration Hazard	×		
	·	Legend: 🗙 – Data either n	not available or does not fill the criteria for classification		

Data available to make classification

## **SECTION 12 Ecological information**

#### Toxicity

Ipoint EC(ECx) 50	t Available Test Duration (hr) 72h 72h	Not Available         Species         Algae or other aquatic plants         Algae or other aquatic plants	Not Available Value 0.1mg/l	Not Available Source
EC(ECx)	72h	Algae or other aquatic plants	0.1mg/l	Source
50				1
	72h	Algae or other aquatic plants		
0		Algae of other aquatic plants	0.69mg/l	4
0	96h	Fish	16.4mg/l	2
50	48h	Crustacea	2mg/l	2
50	96h	Algae or other aquatic plants	2.27mg/l	4
lpoint Te	est Duration (hr)	Species	Value	Source
ilable	ot Available	Not Available	Not Available	Not Available
50 	0 point Te lable Ne ted from 1.1	0 96h point Test Duration (hr) lable Not Available ted from 1. IUCLID Toxicity Data 2. Europe ECHA Reg	0       96h       Algae or other aquatic plants         point       Test Duration (hr)       Species         lable       Not Available       Not Available         ted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information	0     96h     Algae or other aquatic plants     2.27mg/l       point     Test Duration (hr)     Species     Value       Not Available     Not Available     Not

Harmful to aquatic organisms.

DO NOT discharge into sewer or waterways.

## Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
hydrogen peroxide	LOW	LOW
Distilled Water	LOW	LOW

#### **Bioaccumulative potential**

Ingredient	Bioaccumulation
hydrogen peroxide	LOW (LogKOW = -1.571)
Mobility in soil	
Ingredient	Mobility
hydrogen peroxide	LOW (KOC = 14.3)

# **SECTION 13 Disposal considerations**

Waste treatment methods	
Product / Packaging disposal	<ul> <li>Containers may still present a chemical hazard/ danger when empty.</li> <li>Return to supplier for reuse/ recycling if possible.</li> <li>Otherwise:</li> <li>If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill.</li> <li>Where possible retain label warnings and SDS and observe all notices pertaining to the product.</li> <li>Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.</li> <li>A Hierarchy of Controls seems to be common - the user should investigate:</li> <li>Reduction</li> </ul>

▶ Reuse
► Recycling
Disposal (if all else fails)
This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. If it has been contaminated, it may be possible to reclaim the product by filtration, distillation or some other means. Shelf life considerations should also be
applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate.
DO NOT allow wash water from cleaning or process equipment to enter drains.
It may be necessary to collect all wash water for treatment before disposal.
In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.
Where in doubt contact the responsible authority.
For small quantities of oxidising agent:
Cautiously acidify a 3% solution to pH 2 with sulfuric acid.
Gradually add a 50% excess of sodium bisulfite solution with stirring.
Add a further 10% sodium bisulfite.
If no further reaction occurs (as indicated by a rise in temperature) cautiously add more acid.
Recycle wherever possible.
Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.
Treat and neutralise at an approved treatment plant. Treatment should involve: Neutralisation followed by: burial in a land-fill specifically licensed to accept chemical and / or pharmaceutical wastes or Incineration in a licensed apparatus (after admixture with suitable combustible material)
Decontaminate empty containers. Observe all label safeguards until containers are cleaned and destroyed.

# **SECTION 14 Transport information**

# Labels Required

Marine Pollutant	NO
HAZCHEM	2P

## Land transport (ADG)

UN number	2014		
UN proper shipping name	HYDROGEN PEROXIDE, AQUEOUS SOLUTION with not less than 20% but not more than 60% hydrogen peroxide (stabilised as necessary)		
Transport hazard class(es)	Class     5.1       Subrisk     8		
Packing group	ll		
Environmental hazard	Not Applicable		
Special precautions for user	Special provisions     Not Applicable       Limited quantity     1 L		

## Air transport (ICAO-IATA / DGR)

	, 			
UN number	2014			
UN proper shipping name	Hydrogen peroxide, aqueous solution with more than 40% but 60% or less hydrogen peroxide (stabilized as necessary); Hydrogen peroxide, aqueous solution with 20% or more but 40% or less hydrogen peroxide (stabilized as necessary)			
	ICAO/IATA Class	5.1		
Transport hazard class(es)	ICAO / IATA Subrisk	8		
	ERG Code	5C		
Packing group	II			
Environmental hazard	Not Applicable			
	Special provisions		A2 A75	
	Cargo Only Packing Instructions		554; Forbidden	
	Cargo Only Maximum Qty / Pack		5 L; Forbidden	
Special precautions for user	Passenger and Cargo Packing Instructions		550; Forbidden	
	Passenger and Cargo Maximum Qty / Pack		1 L; Forbidden	
	Passenger and Cargo Limited Quantity Packing Instructions		Y540; Forbidden	
	Passenger and Cargo Limited Maximum Qty / Pack		0.5 L; Forbidden	

## Sea transport (IMDG-Code / GGVSee)

UN number	2014
UN proper shipping name	HYDROGEN PEROXIDE, AQUEOUS SOLUTION with not less than 20% but not more than 60% hydrogen peroxide (stabilized as necessary)

Continued...

#### Hydrogen peroxide 30% (100 vol)

Transport hazard class(es)	IMDG Class	_	
	IMDG Subrisk		
Packing group	П		
Environmental hazard	Not Applicable		
Special precautions for user	EMS Number Special provisions Limited Quantities	F-H , S-Q Not Applicable	

Transport in bulk according to Annex II of MARPOL and the IBC code Not Applicable

#### Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
hydrogen peroxide	Not Available
Distilled Water	Not Available
<b>-</b>	

#### Transport in bulk in accordance with the ICG Code

Product name	Ship Type
hydrogen peroxide	Not Available
Distilled Water	Not Available

# **SECTION 15 Regulatory information**

#### Safety, health and environmental regulations / legislation specific for the substance or mixture

## hydrogen peroxide is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) -Schedule 5 Australia Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule  $\mathbf{6}$ 

Australian Inventory of Industrial Chemicals (AIIC)

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

#### Distilled Water is found on the following regulatory lists

Australian Inventory of Industrial Chemicals (AIIC)

#### **National Inventory Status**

National Inventory	Status	
Australia - AIIC / Australia Non-Industrial Use	Yes	
Canada - DSL	Yes	
Canada - NDSL	No (hydrogen peroxide; Distilled Water)	
China - IECSC	Yes	
Europe - EINEC / ELINCS / NLP	Yes	
Japan - ENCS	Yes	
Korea - KECI	Yes	
New Zealand - NZIoC	Yes	
Philippines - PICCS	Yes	
USA - TSCA	Yes	
Taiwan - TCSI	Yes	
Mexico - INSQ	Yes	
Vietnam - NCI	Yes	
Russia - FBEPH	Yes	
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)	

## **SECTION 16 Other information**

Revision Date	30/12/2020
Initial Date	23/10/2009

#### **SDS Version Summary**

Version	Date of Update	Sections Updated
4.1.1.1	01/11/2019	One-off system update. NOTE: This may or may not change the GHS classification
5.1.1.1	30/12/2020	Classification change due to full database hazard calculation/update.
5.1.2.1	26/04/2021	Regulation Change

	Date of Update	Sections Updated
5.1.3.1	03/05/2021	Regulation Change
5.1.4.1	06/05/2021	Regulation Change
5.1.5.1	10/05/2021	Regulation Change
5.1.5.2	30/05/2021	Template Change
5.1.5.3	04/06/2021	Template Change
5.1.5.4	05/06/2021	Template Change
5.1.6.4	07/06/2021	Regulation Change
5.1.6.5	09/06/2021	Template Change
5.1.6.6	11/06/2021	Template Change
5.1.6.7	15/06/2021	Template Change
5.1.7.7	17/06/2021	Regulation Change

#### Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

#### **Definitions and abbreviations**

PC-TWA: Permissible Concentration-Time Weighted Average PC-STEL: Permissible Concentration-Short Term Exposure Limit IARC: International Agency for Research on Cancer ACGIH: American Conference of Governmental Industrial Hygienists STEL: Short Term Exposure Limit TEEL: Temporary Emergency Exposure Limit. IDLH: Immediately Dangerous to Life or Health Concentrations ES: Exposure Standard OSF: Odour Safety Factor NOAEL :No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index AIIC: Australian Inventory of Industrial Chemicals DSL: Domestic Substances List NDSL: Non-Domestic Substances List IECSC: Inventory of Existing Chemical Substance in China EINECS: European INventory of Existing Commercial chemical Substances ELINCS: European List of Notified Chemical Substances NLP: No-Longer Polymers ENCS: Existing and New Chemical Substances Inventory KECI: Korea Existing Chemicals Inventory NZIoC: New Zealand Inventory of Chemicals PICCS: Philippine Inventory of Chemicals and Chemical Substances TSCA: Toxic Substances Control Act TCSI: Taiwan Chemical Substance Inventory INSQ: Inventario Nacional de Sustancias Químicas NCI: National Chemical Inventory FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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